

Isoquinuclidine Mimics of β -D-Glucopyranosides: Differences and Similarities in the Mechanism of Action of some β -D-Glucosidases and a β -D-Mannosidase

by Matthias Böhm, Edwige Lorthiois, Muthuppalaniappan Meyyappan, and Andrea Vasella*

Laboratorium für Organische Chemie, ETH-Hönggerberg, Wolfgang-Pauli Strasse 10, CH-8093 Zürich

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The D-*gluco*-isoquinuclidines **3** and **4** were prepared and tested as inhibitors of the β -glucosidases from *Caldocellum saccharolyticum* and from sweet almonds; the results are compared to the inhibition of snail β -mannosidase by the D-*manno*-isoquinuclidines **1** and **2**. Exploratory experiments in the racemic series showed that treatment of the ester epoxide **6** with benzyl alcoholates leads only to epimerisation, transesterification, and formation of the cyclopropane **9**. Ring opening of the reduced epoxide **13** by NaN_3 proceeded regioselectively to provide **14**. Treatment of the C(6)–O-triflate **16** with AcOCs induced a rearrangement; the reaction with NaN_3 gave the C(5)-azido derivative **14**. The acetoxy triflate **18**, however, reacted with AcOCs to provide the desired *gluco*-isoquinuclidine **19**. Similarly, the enantiomerically pure acetoxy triflate **22** provided the D-*gluco*-isoquinuclidine **24**, which was reduced and deprotected to provide **3** and **4**. The deoxy analogues **30** and **31** were obtained by reductive deiodination of the iodide **27**, derived from **22**. The D-*gluco*-isoquinuclidines **3**, **4**, **30**, and **31** are much weaker inhibitors of β -glucosidases than the D-*manno*-analogues **1** and **2** of snail β -mannosidase. The *N*-benzyl derivative **3** is a weaker inhibitor than the *N*-unsubstituted analogue in the *gluco*-series, while it is a much stronger inhibitor in the *manno*-series. A consideration of the $\text{p}K_{\text{HA}}$ values of the isoquinuclidines **1–4** and the pH value of the enzyme assays suggests that the D-*gluco*-isoquinuclidines are poor mimics of the shape of a reactive, enzyme-bound *gluco*-conformer, while the D-*manno*-analogues are reasonably good mimics of a reactive, enzyme-bound *manno*-conformer. The inhibition results may also suggest that the glycosidase induced lengthening of the scissile bond and rehybridisation of the anomeric centre are more strongly correlated with the change of the ground-state conformation during hydrolysis of β -D-glucopyranosides than of β -D-mannopyranosides.

Introduction. – Stereoelectronic control of glycoside hydrolysis requires a coplanar arrangement of the scissile bond and a doubly occupied, nonbonding orbital of the ring O-atom, *i.e.*, a pseudoaxial orientation of the scissile bond [1]. β -D-Glycopyranosides possessing an equatorial glycosidic bond must, therefore, change conformation to undergo hydrolysis. This aspect and the cationic character of the transition state are expected to be common to all β -D-glycopyranosidases¹⁾. One must, however, expect differences for β -D-glycosidases belonging to different families²⁾ and hydrolyzing different substrates. Glycosidase inhibitors³⁾ may contribute to an analysis of similarities and differences of the mechanism of action of individual β -glycosidases. Recently, we became interested in the difference between the mechanism of action of

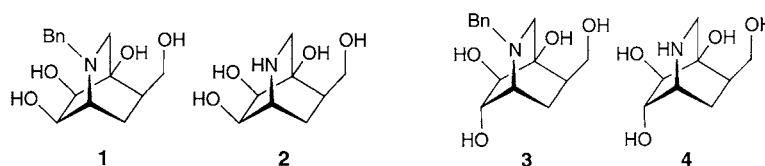
1) For review articles on the mechanism of glycoside hydrolysis, see [2–4].

2) A regularly updated database with over 6000 glycosidases classified in 89 families [5] is available on the internet (<http://afmb.cnrs-mrs.fr/CAZY/>).

3) For review articles on glycosidase inhibitors, see [4][6].

β -D-glucosidases and snail β -D-mannosidase. The catalytic nucleophile of retaining β -D-glucosidases of family 1 [7] and 3 [8] strongly interacts with the C(2)OH group of the substrate, an interaction that is geometrically not feasible for β -D-mannosidases. We speculated about the effect of this interaction [2], concluding that the mechanism of action of β -D-glucosidases and β -D-mannosidases should differ at the beginning of the reaction, but lead to similar cationic intermediates. Not too surprisingly then, a comparison of the inhibition by *gluco*- and *manno*-configured tetrahydroimidazopyridine-type inhibitors did not point to differences between the mechanism of action of the β -glucosidases from *Caldocellum saccharolyticum* (family 1) and from sweet almonds [9], and the β -D-mannosidase from snail [10]. These inhibitors mimic shape and charge of an oxycarbenium cation, but also the cooperative interaction of the substrate with the catalytic acid and nucleophile [11] that is typical for an early stage of the hydrolysis. Inhibitors mimicking the shape of an early stage of the reaction may, however, evidence differences in the mechanism of action of these glycosidases not visible for inhibitors with a trigonal anomeric centre.

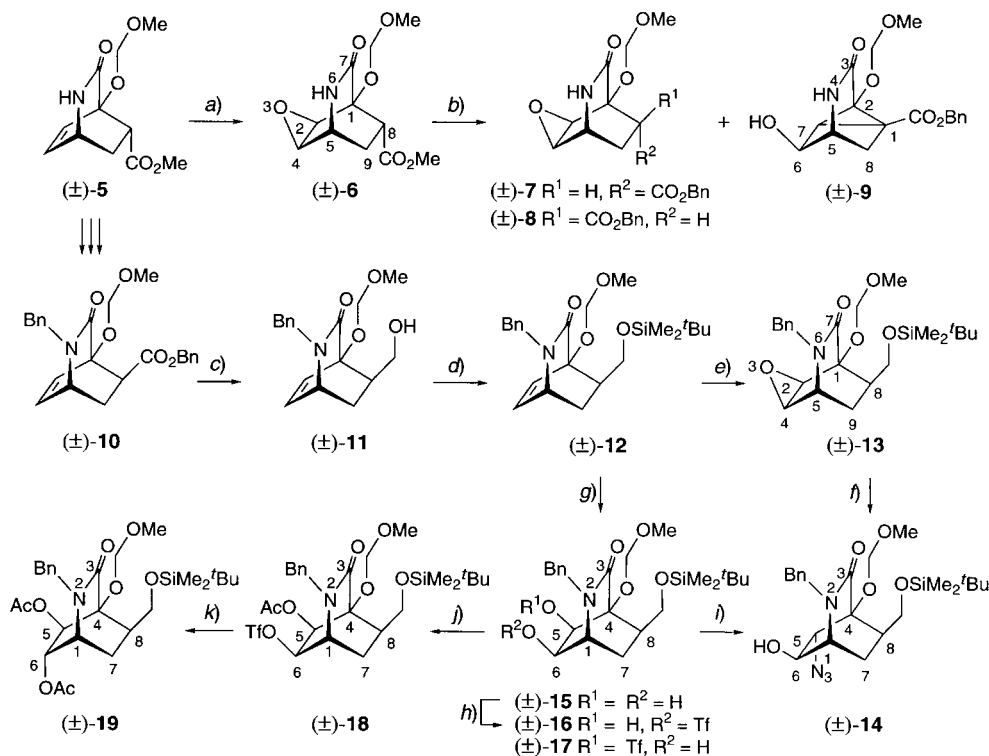
We found that the *manno*-configured isoquinuclidines **1** and **2** [12] mimicking the ^{1,4}B conformer of a β -D-mannopyranoside are selective inhibitors of snail β -mannosidase (at pH 4.5: $K_i = 0.001$ mM, $\alpha = 1.9$ for **1**; $K_i = 1.2$ mM, $\alpha = 1.1$ for **2**), the strength and type of inhibition depending on the pH value⁴). An evaluation of the main factors contributing to the inhibition (shape, pK_{HA} value, and *N*-substituent) suggested that the inhibition results agree with the hypothesis of a conformational change of the substrate, as discussed above. We wondered if the analogous D-*gluco*-isoquinuclidines **3** and **4** would similarly inhibit β -glucosidases, or if this inhibition would differ from the one of snail β -mannosidase by the D-*manno*-isoquinuclidines **1** and **2**. In the following, we describe the synthesis and inhibitory effect of the D-*gluco*-isoquinuclidines **3** and **4** on several glucosidases and on snail β -mannosidase, and compare the inhibition of glucosidases by the D-*gluco*-isoquinuclidines **3** and **4** to the inhibition of snail β -mannosidase by the D-*manno*-isoquinuclidines **1** and **2**.



Synthesis. – We have recently described an efficient synthesis of the D-*manno*-configured isoquinuclidines **1** and **2** via the alkene **20** (cf. Scheme 2). We planned to synthesise the D-*gluco*-configured isoquinuclidines **3** and **4** by a regioselective ring opening of an epoxide derived from **20**, or by an inverting substitution of the *cis*-diol **21**. Exploratory experiments were performed with the racemic alkene **5** [12][14] (Scheme 1). Epoxidation of **5** with *meta*-chloroperbenzoic acid (*m*CPBA) in the presence of 4,4'-thiobis[6-(*tert*-butyl)-3-methylphenol] [15] yielded 85% of the epoxide

⁴) A $B_{2,5}$ conformer has been postulated by Davies and co-workers [13] as reactive conformation of the substrate of the *Pseudomonas cellulosa* mannanase Man26A.

Scheme 1



a) *m*CPBA, 4,4'-thiobis[6-(*tert*-butyl)-3-methylphenol], $ClCH_2CH_2Cl$, Δ ; 85%. *b*) CsF, BnOH, DMF, Δ ; ((\pm)-7/((\pm)-8 7:3 (33%) and (\pm)-9 (10%). *c*) $LiBH_4$, THF, Δ ; 78%. *d*) 1. MeLi, THF, 0° ; 2. t -BuMe₂SiCl, THF, 0° ; 87%. *e*) Dimethyldioxirane, acetone, 0° ; 71%. *f*) NaN_3 , DMF, Δ ; (\pm)-13 (44%) and (\pm)-14 (20%). *g*) *N*-Methylmorpholine *N*-oxide monohydrate (NMO·H₂O), OsO₄, acetone/H₂O; 81%. *h*) (CF₃SO₂)₂O (Tf₂O), pyridine, -10° ; (\pm)-16 (81%). *i*) NaN_3 , DMF, Δ ; 78%. *j*) 1. Tf₂O, pyridine; 2. Ac₂O, pyridine; 66%. *k*) AcOCs, DMF, Δ ; 84%.

6⁵). Its structure was established by X-ray crystal-structure analysis⁶) (*Fig.*). Treatment of **6** with BnOH in the presence of CsF gave the epimeric benzyl esters **7/8** 7:3 (33%) and the cyclopropane **9** (10%). No benzyl ether was observed. BnOLi and BnONa led only to transesterification and epimerisation. We, therefore, transformed the ester **5** into the silyl ether **12** before epoxidation or dihydroxylation. Dealkylative epimerisation of the ester **5**, followed by benzylation, provided the benzyl ester **10** (46% [12]). Its reduction with $LiBH_4$ in boiling THF yielded 78% of the alcohol **11**. Silylation gave the silyl ether **12** (87%). Its oxidation with dimethyldioxirane led to the epoxide **13** (71%). All attempts to substitute **13** with AcOCs failed to provide the

⁵) For a similar diastereoselective epoxidation of an isoquinuclidine lactam, see [16].

⁶) The crystallographic data have been deposited with the *Cambridge Crystallographic Data Centre* as deposition No. CCDC-217684. Copies of the data can be obtained, free of charge, on application to the CCDC, 12 Union Road, Cambridge CB2 1EZ, UK (fax: +44(1223)336033; e-mail: deposit@ccdc.cam.ac.uk).

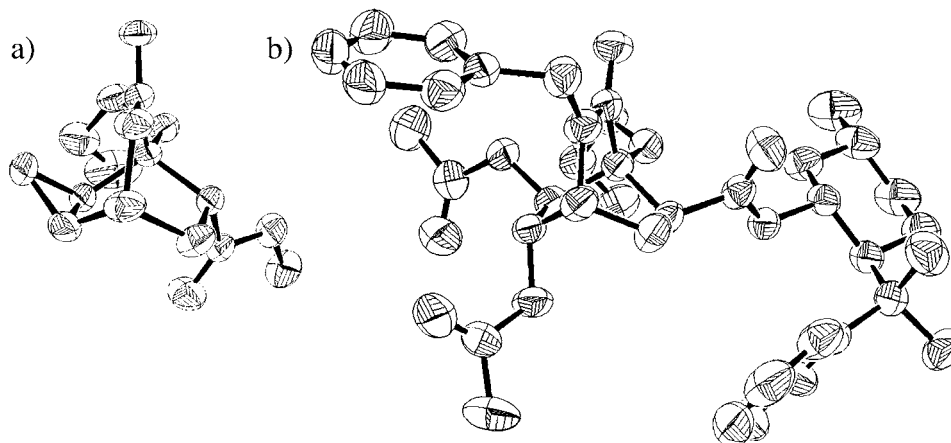


Figure. ORTEP Representation of the Crystal Structure a) of the Epoxide (\pm)-**6** and b) of the Diacetoxy 8-Phenylmenthyl Ester **24**.

desired *trans*-acetoxy alcohol, and treatment of **13** with NaN_3 gave only the azido alcohol **14** (20%). The regioselectivity of the ring opening shows that the effect of the electronegative C(5)-*N*-substituent is more important than the combined effects of the (unfavourably oriented) C(1)-alkoxy group and of the steric hindrance of C(2).

Dihydroxylation of the alkene **12** led diastereoselectively to the diol **15** (81%). It was *O*-sulfonylated with trifluoromethanesulfonic acid anhydride (Tf_2O) in pyridine at -10° to give regioselectively the triflate **16** (81%). Treatment of **16** with AcOCs in DMF at 120° did not lead to substitution; IR and $^1\text{H-NMR}$ analysis of the crude product suggested formation of a deoxy ketone (*cf.* [17]). Surprisingly, treatment of **16** with NaN_3 in DMF at 85° led to substitution at C(5), providing the azido alcohol **14** (78%). This result suggests a migration of the Tf group from *O*-C(6) to *O*-C(5) and a more facile substitution of the resulting triflate **17**. Acetylation of *O*-C(5) should prevent this migration and allow substitution at C(6). The acetoxy triflate **18** was prepared by treating the diol **15** with Tf_2O in pyridine and then with Ac_2O (66%). Substitution of **18** with AcOCs in DMF at 60° gave indeed 84% of the desired *gluco*-configured diacetate **19**.

The configurations of the epoxides **6** and **13** were deduced from the coupling between H-C(4) and the bridgehead H-C(5) ($J(4,5) = 4.4$ and 4.5 Hz, resp.) and from the *W*-coupling between H-C(4) and H_{endo} -C(9) ($J(4,6_{\text{endo}}) = 1.3$ and < 2.2 Hz, resp.; Table 2 in *Exper. Part*). This interpretation is in keeping with the crystal structure of **6**. The structure of the cyclopropane **9** agrees with the multiplicity and chemical shifts of the signals for C(1) (*s* at 32.01 ppm), C(2) (*s* at 68.18 ppm), and C(7) (*d* at 37.54 ppm). The constitution of the azido alcohol **14** was assigned on the basis of a decoupling experiment; irradiation of the H-C(6) signal affected the signals for H-C(1) ($J(1,6) = 2.1$), H-C(5) ($J(5,6) = 0.8$), and OH ($J(6,\text{OH}) = 6.2$ Hz). The configuration of **14** was deduced from the small coupling between H-C(5) and H-C(6) ($J(5,6) = 0.8$ Hz) and by comparing the coupling of H-C(6) with the bridgehead H-C(1) ($J(1,6) = 2.1$ Hz) to the coupling of H-C(1) with H_{endo} -C(7) and

$H_{exo}-C(7)$ ($J(1,7_{endo})=2.5$, $J(1,7_{exo})=4.4$ Hz). This deduction was corroborated by NOE experiments confirming that $H-C(6)$, $H_{endo}-C(7)$, and $H-C(8)$ are on the same, and $H-C(5)$ and $H_{endo}-C(7)$ on opposite sides of the cyclohexane ring of **14**.

The configuration of the diol **15** was deduced by comparing the coupling of $H-C(6)$ with the bridgehead $H-C(1)$ ($J(1,6)=1.9$ Hz) to the coupling of $H-C(1)$ with $H_{endo}-C(7)$ and $H_{exo}-C(7)$ ($J(1,6_{endo})=2.2$; $J(1,6_{exo})=3.7$ Hz). This deduction was corroborated by NOE experiments confirming that $H-C(6)$, $H_{endo}-C(7)$, and $H-C(8)$ are on the same side of the cyclohexane ring of **15**. The constitution of the triflate **16** was assigned on the basis of decoupling experiments confirming the coupling between OH and $H-C(5)$ ($J(5,OH)=3.7$ Hz), $H-C(5)$ and $H-C(6)$ ($J(5,6)=7.9$ Hz), and $H-C(6)$ and $H-C(1)$ ($J(1,6)=1.7$ Hz). The configuration of the *gluco*-configured diacetate **19** was deduced by comparing the coupling of $H-C(6)$ with the bridgehead $H-C(1)$ ($J(1,6)=4.4$ Hz) to the coupling of $H-C(1)$ with $H_{endo}-C(7)$ and $H_{exo}-C(7)$ ($J(1,7_{endo})=2.2$, $J(1,7_{exo})=3.0$ Hz); as expected, the coupling between $H-C(5)$ and $H-C(6)$ is small ($J(5,6)=1.6$ Hz). Also this deduction was corroborated by NOE experiments, confirming that $H-C(5)$ and $H-C(8)$ are on the same, and $H-C(6)$ and $H-C(8)$ on opposite sides of the cyclohexane ring of **19**.

On the basis of these results, we treated the enantiomerically pure diol **21** with Tf_2O and then with Ac_2O in pyridine to yield the acetoxy triflate **22** (81%) and the bis-triflate **23** (11%; *Scheme 2*). The crystalline diol **21** was obtained in 77% yield by dihydroxylation of the alkene **20** [12]. Substitution of **22** with AcOCs yielded 94% of the *D-gluco*-configured diacetate **24**. Reduction of **24** ($LiAlH_4$) and acetylation of the crude product gave the *O*-acetylated *N*-benzyl amine **25** (88%). It was deprotected by cleaving the methoxymethyl acetal with Me_3SiBr , followed by deacetylation, to yield 87% of the *N*-benzyl-*D-gluco*-isoquinuclidine **3**. Catalytic hydrogenolysis of **3** under acidic conditions led to the *N*-unsubstituted *D-gluco*-isoquinuclidine **4** (95%). This synthesis provided the enantiomerically pure *N*-unsubstituted *D-gluco*-isoquinuclidine **4** in eight steps and 55% yield from the diol **21**.

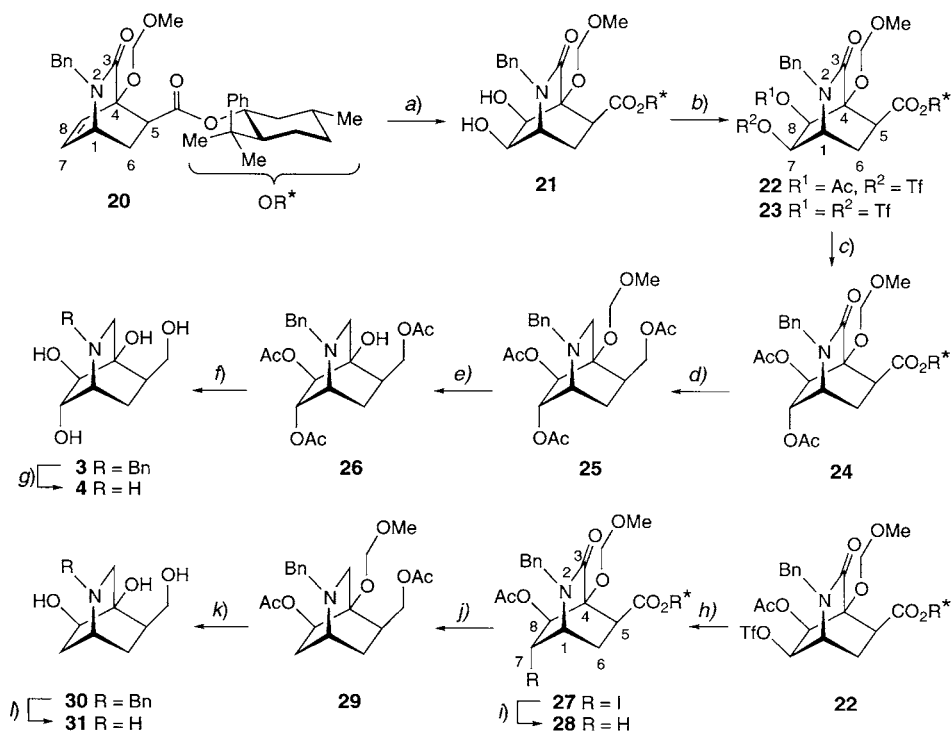
The X-ray crystal-structure analysis of the enantiomerically pure diacetoxy 8-phenylmenthyl ester **24**⁷⁾ (*Fig.*) established the *D-gluco*-configuration of **24** and thereby also of **3** and **4**. Selected chemical shifts and coupling constants of the isoquinuclidines **3**, **4**, and **22–25** are given in *Table 3 (Exper. Part)*.

To study the interaction of glucosidases with the isoquinuclidine OH group mimicking C(2)OH of the substrate, we also synthesised the 2-deoxy-*D-gluco*-isoquinuclidines **30** and **31**⁸⁾. Treatment of the acetoxy triflate **22** with Bu_4NI gave the iodide **27** (95%) that was deiodinated with Bu_3SnH in toluene at 43° to provide the lactam **28** in 90% yield. Selected chemical shifts and coupling constants of **27** and **28** are given in *Table 3 (Exper. Part)*. Reduction of **28** with $LiAlH_4$, followed by acetylation of the crude product, gave the *O*-acetylated *N*-benzyl amine **29** (91%) that was deprotected by treatment with ethanolic HCl to yield 89% of the *N*-benzyl-2-deoxy-

7) The crystallographic data have been deposited with the *Cambridge Crystallographic Data Centre* as deposition No. CCDC-217685. Copies of the data can be obtained, free of charge, on application to the CCDC, 12 Union Road, Cambridge CB2 1EZ, UK (fax: +44(1223)336033; e-mail: deposit@ccdc.cam.ac.uk).

8) The isoquinuclidines **30** and **31** were prepared by *Corinne Baumgartner* as part of her undergraduate training.

Scheme 2



a) $\text{NMO} \cdot \text{H}_2\text{O}$, OsO_4 , THF/acetone/ H_2O ; 77%. b) 1. Tf_2O , pyridine; 2. Ac_2O , pyridine; **22** (81%) and **23** (11%).
 c) CsOAc , DMF, Δ ; 94%. d) 1. LiAlH_4 , THF, Δ ; 2. 4-(Dimethylamino)pyridine (DMAP), Ac_2O , pyridine; 88%. e) Me_3SiBr , molecular sieves (4 Å), CH_2Cl_2 , Δ ; 88%. f) NH_3/MeOH ; 99%. g) H_2 , $\text{Pd}(\text{OH})_2\text{-C}$, $\text{MeOH}/\text{H}_2\text{O}/\text{HCl}$; 95%. h) Bu_4NI , DMF, Δ ; 95%. i) 2,2'-Azobis(isobutyronitrile) (AIBN), Bu_3SnH , toluene, Δ ; 90%. j) 1. LiAlH_4 , THF, Δ ; 2. DMAP, Ac_2O , pyridine; 91%. k) $\text{EtOH}/\text{H}_2\text{O}/\text{HCl}$; 89%. l) H_2 , $\text{Pd}(\text{OH})_2/\text{C}$, $\text{MeOH}/\text{H}_2\text{O}/\text{HCl}$; 92%.

D-gluco-isoquinuclidine **30**. Catalytic hydrogenolysis of **30** under acidic conditions led to the N-unsubstituted 2-deoxy-D-gluco-isoquinuclidine **31** (92%). This synthesis provided **31** in eight steps and 52% yield from the diol **21**.

Inhibition studies. – The D-gluco-isoquinuclidines **3** and **4** are both weak inhibitors of the β -glucosidases from *C. saccharolyticum* and from sweet almonds, and of the cellulase Cel7A from *Trichoderma reesei*⁹⁾; the N-unsubstituted D-gluco-isoquinuclidine **4** is a weak inhibitor of the β -D-glucan glucohydrolase from barley, whereas the N-benzyl D-gluco-isoquinuclidine **3** did not inhibit this enzyme¹⁰⁾ (Table 1). The D-gluco-

⁹⁾ We thank Prof. Dr. Jerry Ståhlberg, Swedish University of Agricultural Sciences, Uppsala, for a generous sample of *T. reesei* Cel7A.

¹⁰⁾ Compound **3** did not inhibit barley β -D-glucan glucohydrolase up to a concentration of 4 mM, when it started to precipitate. We thank Dr. M. Hrmova, University of Adelaide, for testing **3** and **4** as barley β -D-glucan glucohydrolase inhibitors.

isoquinuclidines **3** and **4** do neither inhibit the *endo*-1,3-glucanase from soybean¹¹⁾ nor the endoglucanase NtEg1 from *Nasutitermes takasagoensis*¹²⁾. Surprisingly, the *N*-benzyl-D-*gluco*-isoquinuclidine **3** is a stronger inhibitor of snail β -mannosidase than of the glucosidases from *C. saccharolyticum* and sweet almonds. The 2-deoxy-D-*gluco*-isoquinuclidines **30** and **31** are weak inhibitors of the glucosidase from *C. saccharolyticum*; the β -glucosidases from sweet almonds are neither inhibited by **30** nor by **31** (Table 1).

Table 1. K_M and IC_{50} Values [mM] for the Inhibition of *C. saccharolyticum* and Sweet Almond β -Glucosidases, Barley β -D-Glucan Glucohydrolase, *T. reesei* Cellulase Cel7A, and Snail β -Mannosidase by the Isoquinuclidines **3**, **4**, **30**, and **31** at the Given Temperature and pH Value

Enzyme	Temp.	pH	K_M	IC_{50} Values			
				3	4	30	31
<i>C. saccharolyticum</i> β -glucosidase	55°	6.80	0.80	2.1 ^{a)}	1.9 ^{b)}	9.4	4.8
Sweet almond β -glucosidases	37°	6.80	3.4	3.3	2.7	n.i. ^{c)}	n.i. ^{c)}
Barley β -D-glucan glucohydrolase	37°	5.25	1.7	n.i. ^{c)}	2.2 ^{d)}	–	–
<i>T. reesei</i> Cel7A	30°	5.72	0.39	6.0	12.6	–	–
Snail β -mannosidase	25°	4.50	0.58	0.15	2.0	–	–

^{a)} $K_i = 1.5$ mM ($\alpha = 5.7$). ^{b)} $K_i = 1.3$ mM ($\alpha = 9.0$). ^{c)} n.i. = no inhibition up to a concentration of [**3**] = 4 mM.

^{d)} $K_i = 2$. ^{e)} n.i. = no inhibition up to a concentration of [**30**] = [**31**] = 4 mM.

To compare the inhibition of the *C. saccharolyticum* and sweet almond β -glucosidases at pH 6.8 by the D-*gluco*-isoquinuclidines **3** ($pK_{HA} = 6.6$) and **4** ($pK_{HA} = 7.8$) with the inhibition of snail β -mannosidase at pH 4.5 by the D-*manno*-isoquinuclidines **1** ($pK_{HA} = 7.5$) and **2** ($pK_{HA} = 8.4$) [12], we considered the difference between the pK_{HA} value of the isoquinuclidines **3** and **4**, and the pH value of the enzyme assays, as we did for **1** and **2** [12]. On the basis of the pH dependence of the inhibition of snail β -mannosidase by the D-*manno*-isoquinuclidines **1** and **2**, we concluded that the amines rather than the ammonium salts act as inhibitors. An extrapolation of the pH-dependent inhibition constants (at pH 4.5: $K_i = 1.0$ μ M, $\alpha = 1.9$ for **1** and $K_i = 1.2$ mM, $\alpha = 1.1$ for **2**) suggested that D-*manno*-isoquinuclidines corresponding to **1** and **2**, but with a (hypothetical) pK_{HA} value equivalent to the pH optimum of the β -mannosidase (pH *ca.* 4.5) would show an inhibition constant of *ca.* 2 nM and *ca.* 2 μ M, respectively. The assumption of a similar dependence on the pH of the inhibition constants for the β -glucosidase from *C. saccharolyticum* by the D-*gluco*-isoquinuclidines **3** and **4** suggests that D-*gluco*-isoquinuclidines with a (hypothetical) pK_{HA} value equivalent to the pH optimum of the β -glucosidase (pH *ca.* 6.8) would show inhibition constants of *ca.* 2 mM and *ca.* 0.3 mM, respectively, *i.e.*, larger or similar to the Michaelis–Menten constant ($K_M = 0.60$ mM). The same assumption suggests IC_{50} values of *ca.* 5 mM and *ca.* 0.5 mM for the inhibition of the β -glucosidases from sweet almonds by D-*gluco*-isoquinuclidines

¹¹⁾ We thank Prof. Dr. J. Ebel, Ludwig-Maximilian-Universität, München, for testing **3** and **4** as inhibitors of the β -D-glucanase from soybean.

¹²⁾ We thank Prof. E. Meyer, Texas A&M University, for a generous sample of the endoglucanase NtEg1 from the termite *N. takasagoensis*.

corresponding to **3** and **4**, but with a (hypothetical) pK_{HA} value equivalent to the pH optimum of the β -glucosidase (pH *ca.* 6.8).

These considerations confirm that isoquinuclidines are much stronger inhibitors of snail β -mannosidase than of the β -glucosidases from *C. saccharolyticum* and from sweet almonds. They evidence that shape, and not charge, is at the origin of this difference¹³). The enzymatic hydrolysis of β -D-mannopyranosides may proceed *via* a conformer that is mimicked (more or less well, as indicated by the mixed-type inhibition) by the D-*manno*-isoquinuclidines **1** and **2**, while the enzymatic hydrolysis of the β -D-glucopyranosides proceeds *via* a significantly different conformer. The weakening effect of the *N*-Bn substituent on the inhibition by the *gluco*-isoquinuclidines **3** and **4** is significant, since the β -glucosidases from sweet almonds show a strong affinity for hydrophobic aglyca [9][19][20]. The destabilising interaction with the *N*-Bn group contrasts with its effect on the inhibition of the β -mannosidase. This may reflect a location of the *N*-Bn group of **3**, as imposed by the inappropriate structure of the D-*gluco*-isoquinuclidine, that significantly deviates from the location of the aglycon part of the substrate on the way to the transition state. One also has to consider the possibility that the change of the ground-state conformation precedes lengthening of the scissile bond and rehybridisation of the anomeric centre during hydrolysis of β -D-mannopyranosides, but not of β -D-glucopyranosides; an incipient conformational change that is strongly correlated with bond breaking would also lead to structures not mimicked by an isoquinuclidine. This, too, would constitute a significant difference to the mechanism of action of snail β -mannosidase.

We thank Dr. B. Schweizer for the determination of the X-ray crystal structures, D. Manser and M. Schneider for the pK_{HA} determinations, Dr. B. Bernet for checking the *Exper. Part*, and the Swiss National Science Foundation and Oxford Glycosciences Ltd., Abingdon, UK, for generous support.

Experimental Part

General. Solvents were distilled before use. If not specified otherwise, all reactions were carried out under a N_2 atmosphere. Normal workup implies pouring the reaction mixture into the indicated sat. aq. soln., extracting into the mentioned org. solvent, if necessary washing with the indicated sat. aq. soln., drying of the org. layer (Na_2SO_4 or $MgSO_4$), filtration, and evaporation of the volatiles. TLC: Merck silica-gel 60F-254 plates; detection with I_2 , by heating with *Mostain* (400 ml of 10% H_2SO_4 soln., 20 g of $(NH_4)_6Mo_7O_{24} \cdot 6 H_2O$, 0.4 g of $Ce(SO_4)_2$) or with $KMnO_4$ soln. (3 g of $KMnO_4$, 20 g of K_2CO_3 , 0.25 ml of AcOH in 300 ml of H_2O). Flash chromatography (FC): silica gel *Fluka 60* (0.04–0.063 mm) if not indicated otherwise. IR Spectra: 0.5% KBr suspension, 3% $CHCl_3$ or CH_2Cl_2 soln. 1H - (300 MHz, if not indicated otherwise) and ^{13}C -NMR (75 MHz, if not indicated otherwise): δ in ppm and J in Hz.

Methyl rac-(1R,2R,4R,5R,8R)-1-(Methoxymethoxy)-7-oxo-3-oxa-6-azatricyclo[3.2.2.0^{2,4}]octane-8-carboxylate ((±)-6). A soln. of (±)-**5** (1.60 g, 6.63 mmol), *m*CPBA (2.86 g, 16.6 mmol), and 4,4'-thiobis[6-(*tert*-butyl)-3-methylphenol] (42 mg, 0.117 mmol) in $ClCH_2CH_2Cl$ (50 ml) was heated to reflux for 4.5 h. Normal workup (CH_2Cl_2/Na_2SO_3 soln., $MgSO_4$) and FC (cyclohexane/AcOEt/MeOH 1:1:0.1) gave (±)-**6** (1.46 g, 85%). White solid. R_f (AcOEt/hexane/MeOH 1:1:0.1) 0.29. M.p. 122° (CH_2Cl_2/Et_2O). IR ($CHCl_3$): 3424w, 3223m, 3007m, 2901w, 1706s, 1437m, 1364m, 1271m, 1175m, 1152m, 1096w, 1060m, 984m, 908w. 1H -NMR ($CDCl_3$): see Table 2; additionally, 5.86 (br. *d*, $J \approx 0.9$, HN); 5.16, 4.94 (*dd*, $J = 7.5$, OCH_2O); 3.72 (*s*, CO_2Me); 3.39 (*s*, $MeOCH_2$); 2.94 (*dd*, $J = 10.3$, 5.8, H–C(8)); 2.33 (*ddd*, $J = 13.2$, 10.3, 4.1, H_{exo} –C(9)); 1.87 (*ddt*, $J = 13.2$, 5.8, 1.3, H_{endo} –C(9)). ^{13}C -NMR ($CDCl_3$): 173.18, 171.06 (2s, 2 C=O); 95.05 (*t*, OCH_2O); 81.76 (*s*, C(1)); 56.95 (*q*, $MeOCH_2$); 52.58

¹³) We consider the possibility of an unfavourable interaction of $CH_2(3)$ of the D-*gluco*-isoquinuclidines **3** and **4** (corresponding to an axial C(4)-substituent in a β -D-glucoside) with the β -glucosidase from *C. saccharolyticum* as improbable, since this enzyme cleaves both β -D-glucosides and β -D-galactosides [18].

Table 2. Selected $^1\text{H-NMR}$ Chemical Shifts [ppm] and Coupling Constants [Hz] of the Racemic Isoquinuclidines (\pm)-**6** to (\pm)-**9**, (\pm)-**13** to (\pm)-**16**, (\pm)-**18**, and (\pm)-**19**. For an easier comparison, all isoquinuclidines are numbered in the same way as (\pm)-**14** to (\pm)-**19**.

	(\pm)- 6	(\pm)- 7	(\pm)- 8	(\pm)- 9	(\pm)- 13	(\pm)- 14	(\pm)- 15	(\pm)- 16	(\pm)- 18	(\pm)- 19
H–C(1)	3.90	3.90	3.90	3.54	3.66	2.80	3.46	3.63	3.71	3.62
H–C(5)	4.06	4.09	3.78	2.88	3.69	3.82	3.97	4.08	5.34	5.05
H–C(6)	3.60	3.61	3.50	4.27	3.40	3.28	3.95	4.98	4.98	4.52
$J(1,5)$	0.6	0.9	^{a)}	–	0.8	–	–	–	–	–
$J(1,6)$	4.4	4.3	4.1	3.0	3.4	2.1	1.9	1.7	1.6	4.4
$J(1,7_{endo})$	1.3	^{b)}	2.1	1.2	^{c)}	2.5	2.2	2.2	2.2	2.2
$J(1,7_{exo})$	4.1	4.3	3.1	3.7	^{c)}	4.4	3.7	4.0	4.1	3.0
$J(5,6)$	4.4	4.3	4.1	–	4.5	0.8	8.4	7.9	7.8	1.6
$J(5,\text{OH})$	–	–	–	–	–	–	2.5	3.7	–	–
$J(6,7_{endo})$	1.3	–	–	–	^{c)}	–	–	–	–	–
$J(6,7_{exo})$	–	–	–	–	–	–	–	–	–	1.6
$J(6,\text{OH})$	–	–	–	–	–	6.2	8.4	–	–	–

^{a)} Broad signal; $J(1,5) < 1.5$ Hz is deduced from the width of the signal at the half of the height. ^{b)} Broad signal; $J(1,7_{endo}) < 2.5$ Hz is deduced from the width of the signal at the half of the height. ^{c)} Broad signals; $J(1,7_{endo}) \approx J(1,7_{exo}) \approx 2.9$ Hz, $J(6,7_{endo}) < 2.2$ Hz are deduced from the width of the signal at the half of the height.

(*q*, CO_2Me); 50.06 (*d*, C(5)); 48.91 (*d*, C(8)); 45.58, 44.15 (2*d*, C(2), C(4)); 31.34 (*t*, C(9)). CI-MS: 258 (57, $[M+H]^+$), 226 (95), 214 (14), 197 (29), 153 (19), 45 (100). Anal. calc. for $\text{C}_{11}\text{H}_{15}\text{NO}_6$ (257.24): C 51.36, H 5.88, N 5.44; found C 51.37, H 5.83, N 5.48.

X-Ray Crystal-Structure Analysis of (\pm)-**6**. The crystals were obtained by recrystallisation from $\text{CH}_2\text{Cl}_2/\text{Et}_2\text{O}$. Triclinic, $P1$, $a = 6.9317(12)$ Å, $b = 9.5462(13)$ Å, $c = 9.625(2)$ Å, $\alpha = 105.880(12)^\circ$, $\beta = 91.401(13)^\circ$, $\gamma = 104.468(12)^\circ$, $V = 590.2(2)$ Å³, $Z = 1$, $D_{\text{calc}} = 1.447$ Mg/m³. From a crystal of size $0.40 \times 0.40 \times 0.20$ mm, 1988 reflexions were measured on an *Enraf Nonius* CAD-4 diffractometer (graphite monochromator, CuK_α radiation, $\lambda = 1.54184$ Å) at 293(2) K, by Dr. B. Schweizer (ETH-Zürich). $R = 0.0610$, $R_w = 0.2324$. The structure was solved by direct methods with SHELXS-96 [21]. The non-H-atoms were refined anisotropically with SHELXL-97 [22]. The H-atoms were calculated and included in the structure factor calculation. Drawings of the molecule were made with ORTEP [23].

Benzyl rac-(1R,2R,4R,5R,8R)- and rac-(1R,2R,4R,5R,8S)-1-(Methoxymethoxy)-7-oxo-3-oxa-6-azatricyclo[3.2.2.0^{2,4}]octane-8-carboxylate ((\pm)-**7** and (\pm)-**8**, resp.), and *Benzyl rac-(1R,2S,5R,6S,7S)-6-Hydroxy-2-(methoxymethoxy)-3-oxo-4-azatricyclo[3.2.1.0^{2,7}]octane-1-carboxylate* ((\pm)-**9**). A soln. of (\pm)-**6** (30 mg, 0.12 mmol) and dry CsF (90 mg, 0.58 mmol) in DMF (1 ml) was treated with BnOH (0.06 ml, 0.58 mmol) at 23° and stirred at 100° for 72 h. Normal workup ($\text{Et}_2\text{O}/\text{ice-water}$; MgSO_4) and FC (cyclohexane/ AcOEt/MeOH 1:1:0.1) gave (\pm)-**7**/ \pm -**8** 7:3 (13 mg, 33%) and (\pm)-**9** (4 mg, 10%).

Data for (\pm)-**7**/ \pm -**8** 7:3: R_f ($\text{AcOEt}/\text{hexane}/\text{MeOH}$ 1:1:0.1) 0.33. $^1\text{H-NMR}$ (CDCl_3): see Table 2; additionally, 7.38–7.28 (*m*, 5 H); 6.38 (br. *d*, $J = 5.6$, 0.7 H), 6.00 (br. *s*, 0.3 H) (HN); 5.29, 4.98 (2*d*, $J = 7.5$, 0.6 H), 5.19, 4.93 (2*d*, $J = 7.5$, 1.4 H) (OCH_2O); 5.21, 5.14 (2*d*, $J = 12.1$, 1.4 H), 5.21, 5.10 (2*d*, $J = 12.5$, 0.6 H) (PhCH_2); 3.48 (*s*, 0.9 H), 3.28 (*s*, 2.1 H) (MeO); 3.09 (*dd*, $J = 9.8$, 4.8, 0.3 H), 3.01 (*dd*, $J = 10.4$, 5.9, 0.7 H) (H–C(8)); 2.36 (*ddd*, $J = 13.2$, 10.4, 4.3, 0.7 H), 2.18 (*ddd*, $J = 13.4$, 4.8, 3.1, 0.3 H) (H_{exo} –C(9)); 2.09 (*ddd*, $J = 13.4$, 9.8, 2.1, 0.3 H), 1.88 (br. *dd*, $J = 13.2$, 5.9, $w_{1/2} = 2.5$, 0.7 H) (H_{endo} –C(9)).

Data for (\pm)-**9**: R_f ($\text{AcOEt}/\text{hexane}/\text{MeOH}$ 1:1:0.1) 0.11. IR (CHCl_3): 3426s, 2957s, 2927s, 2871s, 2858s, 1731s, 1685s, 1458s, 1379m, 1359m, 1270m, 1177m, 1167m, 1109w, 1089w, 998w, 978w. $^1\text{H-NMR}$ (CDCl_3): see Table 2; additionally, 7.36–7.30 (*m*, 5 arom. H, HN); 5.20, 5.12 (2*d*, $J = 12.5$, PhCH_2); 4.99, 4.77 (2*d*, $J = 6.9$, OCH_2O); 4.41–4.07 (*m*, exchange with D_2O , HO); 4.27 (addn. of $\text{D}_2\text{O} \rightarrow d$, $J = 3.0$, H–C(6)); 3.33 (*s*, MeO); 2.12 (*dd*, $J = 12.8$, 3.7, H_{exo} –C(8)); 2.00 (*dd*, $J = 12.8$, 1.2, H_{endo} –C(8)). $^{13}\text{C-NMR}$ (CDCl_3): 169.66, 167.43 (2s, 2 C=O); 135.78 (*s*); 128.92 (2*d*); 128.71 (2*d*); 128.58 (*d*); 97.66 (*t*, OCH_2O); 70.04 (*d*, C(6)); 67.55 (*t*, PhCH_2); 68.18 (*s*, C(2)); 57.10 (*q*, MeO); 51.21 (*d*, C(5)); 37.54 (*d*, C(7)); 33.24 (*t*, C(8)); 32.01 (*s*, C(1)). CI-MS: 334 (65, $[M+H]^+$), 302 (48), 184 (10), 91 (100), 45 (30).

rac-(1*R*,4*S*,8*R*)-2-Benzyl-8-(hydroxymethyl)-4-(methoxymethoxy)-2-azabicyclo[2.2.2]oct-5-ene-3-one ((±)-**11**). A boiling soln. of LiBH₄ (207 mg, 9.42 mmol) in THF (4.8 ml) was treated dropwise with a soln. of (±)-**10** (640 mg, 1.57 mmol) in THF (4.8 ml), heated to reflux for 90 min, cooled to 23°, treated dropwise with MeOH (5 ml), and stirred at 23° for 1 h. Normal workup (AcOEt/H₂O; Na₂SO₄) and FC (AcOEt/hexane/MeOH 1:1:0.1) gave (±)-**11** (374 mg, 78%). Colourless oil. *R*_f (AcOEt/hexane/MeOH 1:1:0.1) 0.27. IR (CH₂Cl₂): 3453*m*, 3068*w*, 2932*m*, 1679*s*, 1452*m*, 1421*m*, 1226*w*, 1162*m*, 1057*m*, 993*m*, 919*w*. ¹H-NMR (CDCl₃): 7.34–7.15 (*m*, 5 H); 6.75 (*dd*, *J* = 8.1, 1.9, irradi. at 3.95 → *d*, *J* = 8.1, H–C(5)); 6.29 (*dd*, *J* = 8.1, 5.9, irradi. at 3.95 → *d*, *J* = 8.1, H–C(6)); 5.16, 5.05 (*2d*, *J* = 7.5, OCH₂O); 4.59, 4.40 (*2d*, *J* = 14.8, PhCH₂); 3.95 (*ddt*, *J* = 5.3, 3.5, 1.9, H–C(1)); 3.70–3.56 (*m*, CH₂–C(8)); 3.49 (*s*, MeO); 2.92 (*br. s*, exchange with D₂O, HO); 2.18 (*ddt*, *J* = 10.0, 7.2, 5.0, H–C(8)); 1.70 (*ddd*, *J* = 12.5, 10.3, 2.2, irradi. at 3.95 → *br. dd*, *J* = 12.5, 10.6, H_{endo}–C(7)); 1.27 (*ddd*, *J* = 12.8, 5.0, 3.4, irradi. at 3.95 → *dd*, *J* = 12.1, 4.7, H_{exo}–C(7)). ¹³C-NMR (CDCl₃): 171.64 (*s*, C(3)); 136.84 (*s*); 135.75, 131.92 (*2d*, C(5), C(6)); 129.14 (*2d*); 128.69 (*2d*); 128.15 (*d*); 96.0 (*t*, OCH₂O), 86.98 (*s*, C(4)); 64.95 (*t*, CH₂–C(8)); 56.13 (*q*, MeO), 52.41 (*d*, C(1)); 48.64 (*d*, C(8)); 42.38 (*t*, PhCH₂); 31.27 (*t*, C(7)). CI-MS: 304 (3, [M + H]⁺), 272 (10), 138 (100), 91 (42), 45 (51). Anal. calc. for C₁₇H₂₁NO₄ (303.96): C 67.31, H 6.98, N 4.62; found: C 67.26, H 7.21, N 4.63.

rac-(1*R*,4*S*,8*R*)-2-Benzyl-8-[(tert-butyl)dimethylsilyloxy]methyl-4-(methoxymethoxy)-2-azabicyclo[2.2.2]oct-5-ene-3-one ((±)-**12**). A cooled (0°) soln. of (±)-**11** (285 mg, 0.94 mmol) in THF (5 ml) was treated with a 1.5*M* soln. of MeLi in Et₂O (0.72 ml, 1.15 mmol), stirred for 15 min, treated with a soln. of ^tBuMe₂SiCl (174 mg, 1.15 mmol) in THF (3 ml), and stirred at 0° for 1 h. Normal workup (AcOEt/NH₄Cl soln.; Na₂SO₄) and FC (hexane/AcOEt 3:7) gave (±)-**12** (341 mg, 87%). Colourless oil. *R*_f (AcOEt/hexane 7:3) 0.60. ¹H-NMR (CDCl₃): 7.32–7.16 (*m*, 5 H); 6.70 (*dd*, *J* = 8.1, 1.9, irradi. at 3.94 → *d*, *J* = 8.1, H–C(5)); 6.28 (*dd*, *J* = 8.1, 5.6, irradi. at 3.94 → *d*, *J* = 8.4, H–C(6)); 5.13, 5.01 (*2d*, *J* = 7.2, OCH₂O); 4.57, 4.37 (*2d*, *J* = 14.9, PhCH₂); 4.02 (*dd*, *J* = 9.7, 4.9, irradi. at 2.12 → *d*, *J* = 10.0, CH_a–C(8)); 3.94 (*ddt*, *J* = 5.0, 3.0, 1.9, H–C(1)); 3.47 (*s*, MeO); 3.30 (*t*, *J* = 9.7, irradi. at 2.12 → *dd*, *J* = 9.3, 2.0, CH_b–C(8)); 2.12 (*tt*, *J* ≈ 9.5, 4.3, H–C(8)); 1.70 (*ddd*, *J* = 13.1, 4.7, 3.1, irradi. at 3.94 → *dd*, *J* = 12.8, 4.7, irradi. at 2.12 → change, H_{exo}–C(7)); 1.62 (*ddd*, *J* = 12.5, 10.3, 1.9, irradi. at 3.94 → *dd*, *J* = 12.8, 9.6, irradi. at 2.12 → change, H_{endo}–C(7)); 0.84 (*s*, *t*-Bu); 0.02, 0.00 (*2s*, Me₂Si). ¹³C-NMR (CDCl₃): 171.04 (*s*, C(3)); 136.71 (*s*); 135.43, 131.87 (*2d*, C(5), C(6)); 128.65 (*2d*); 128.20 (*2d*); 127.57 (*d*); 95.17 (*t*, OCH₂O); 84.05 (*s*, C(4)); 63.73 (*t*, CH₂–C(8)); 55.56 (*q*, MeO); 52.44 (*t*, PhCH₂); 48.25 (*d*, C(1)); 42.46 (*d*, C(8)); 31.39 (*t*, C(7)); 25.91 (*q*, Me₃CSi); 18.26 (*s*, Me₃CSi); –5.34, –5.41 (*2q*, Me₂Si). CI-MS: 418 (7, [M + H]⁺), 386 (27), 360 (41), 258 (33), 145 (35), 115 (100), 91 (90), 45 (57). Anal. calc. for C₂₃H₃₅NO₄Si (417.62): C 66.15, H 8.45, N 3.35; found: C 65.14, H 8.45, N 3.42.

rac-(1*R*,2*R*,4*R*,5*R*,8*R*)-6-Benzyl-8-[(tert-butyl)dimethylsilyloxy]methyl-1-(methoxymethoxy)-3-oxa-6-azatricyclo[3.2.2.0^{2,4}]octan-7-one ((±)-**13**). A cooled (0°) soln. of (±)-**12** (44 mg, 0.10 mmol) in CH₂Cl₂ (2 ml) was treated with a freshly prepared *ca.* 0.1*M* soln. of dimethyldioxirane in acetone [24] (2.1 ml, *ca.* 0.21 mmol) and stirred at 0° for 3 h, when TLC showed an incomplete reaction. The mixture was treated with a *ca.* 0.1*M* soln. of dimethyldioxirane in acetone (1.0 ml, *ca.* 0.10 mmol) and stirred at 0° for 20 min. Normal workup (CH₂Cl₂, Na₂S₂O₃ soln.; Na₂SO₄) and FC (cyclohexane/EtOAc 4:1) gave (±)-**13** (32 mg, 71%). *R*_f (cyclohexane/AcOEt 4:1) 0.22. IR (CH₂Cl₂): 2955*s*, 2930*s*, 2857*m*, 1687*s*, 1495*m*, 1463*m*, 1350*w*, 1160*s*, 1106*s*, 1059*s*, 996*s*, 920*w*, 839*s*. ¹H-NMR (200 MHz, CDCl₃): see Table 2; additionally, 7.38–7.26 (*m*, 5 H); 5.25, 5.02 (*2d*, *J* = 7.0, OCH₂O); 4.83, 4.23 (*2d*, *J* = 15.0, PhCH₂); 3.92 (*dd*, *J* = 9.6, 3.6, CH_a–C(8)); 3.48 (*s*, MeO); 3.33 (*t*, *J* ≈ 9.6, CH_b–C(8)); 2.29 (*ddd*, *J* ≈ 8.5, 5.0, 3.7, H–C(8)); 1.81 (*dd*, *J* ≈ 13.8, 10.4, *w*_{1/2} = 2.9, H_{endo}–C(9)); 1.75 (*br. dd*, *J* ≈ 13.7, 5.8, *w*_{1/2} = 2.9, H_{exo}–C(9)); 0.86 (*s*, *t*-Bu); 0.04, 0.02 (*2s*, Me₂Si). CI-MS: 434 (41, [M + H]⁺), 402 (52), 376 (59), 316 (100), 202 (26), 91 (84), 45 (44).

rac-(1*R*,4*R*,5*S*,6*R*,8*R*)-5-Azido-2-benzyl-8-[(tert-butyl)dimethylsilyloxy]methyl-6-hydroxy-4-(methoxymethoxy)-2-azabicyclo[2.2.2]octan-3-one ((±)-**14**). A soln. of (±)-**13** (32 mg, 0.073 mmol) and NaN₃ (12 mg, 0.18 mmol) in DMF (3 ml) was stirred at 120° for 20 h. Normal workup (AcOEt/H₂O; Na₂SO₄) and FC (cyclohexane/AcOEt 7:3) gave (±)-**13** (14 mg, 44%) and (±)-**14** (7 mg, 20%).

Data of (±)-**14**: *R*_f (cyclohexane/AcOEt 13:7) 0.46. IR (CH₂Cl₂): 3595*w*, 2955*s*, 2950*s*, 2856*m*, 2114*s*, 1683*s*, 1462*m*, 1156*m*, 1096*m*, 1044*m*, 994*m*, 840*s*. ¹H-NMR (200 MHz, C₆D₆): see Table 2; additionally, 7.22–7.10 (*m*, 5 H); 5.70 (*d*, *J* = 8.0, irradi. at 3.82 → NOE of 3.1%), 5.00 (*d*, *J* = 8.0) (OCH₂O); 4.84, 4.10 (*2d*, *J* = 14.5, PhCH₂); 4.01 (*dd*, *J* = 9.5, 4.2, CH_a–C(8)); 3.82 (irradi. at 3.28 → NOE of 1.9%, H–C(5)); 3.56 (*t*, *J* ≈ 9.5, CH_b–C(8)); 3.37 (*s*, MeO); 3.28 (irradi. at 3.82 → *dd*, *J* ≈ 4.6, 0.8, irradi. at 2.80 → *br. d*, *J* ≈ 4.6, irradi. at 3.82 → NOE of 2.0%, H–C(6)); 2.80 (irradi. at 3.28 → NOE of 5.7%, H–C(1)); 2.59 (*tt*, *J* ≈ 9.4, 4.4, H–C(8)); 1.54 (*dt*, *J* = 14.1, 4.5, irradi. at 2.80 → *dd*, *J* ≈ 14.1, 4.4, H_{exo}–C(7)); 1.26 (*ddd*, *J* = 14.1, 10.4, 2.5, irradi. at 2.80 → *br. dd*, *J* ≈ 14.1, 10.4, irradi. at 3.28 → NOE of 2.6%, H_{endo}–C(7)); 1.12 (*d*, *J* = 6.2, HO–C(6)); 0.95 (*s*, *t*-Bu); 0.06, 0.04 (*2s*, Me₂Si). ¹³C-NMR (125 MHz, C₆D₆): 168.89 (*s*, C(3)); 137.71 (*s*); 137–125 (several *d*); 94.74

(*t*, OCH₂O); 81.16 (*s*, C(4)); 76.56 (*d*, C(6)); 69.74 (*d*, C(5)); 63.83 (*t*, CH₂–C(8)); 56.18 (*d*, C(1)); 56.02 (*q*, MeO); 49.75 (*t*, PhCH₂); 36.64 (*d*, C(8)); 26.76 (*t*, C(7)); 26.17 (*q*, Me₃CSi); 18.58 (*s*, Me₃CSi); – 5.28, – 5.40 (2*q*, Me₂Si). CI-MS: 477 (6, [M + H]⁺), 449 (25), 417 (38), 391 (61), 359 (86), 202 (34), 91 (100), 75 (28), 45 (19).

rac-(1*R*,4*R*,5*R*,6*R*,8*R*)-2-Benzyl-8-[[*tert*-butyl]dimethylsilyloxy]methyl]-5,6-dihydroxy-4-(methoxymethoxy)-2-azabicyclo[2.2.2]octan-3-one ((±)-**15**). A soln. of (±)-**12** (640 mg, 1.53 mmol) and *N*-methylmorpholine *N*-oxide · H₂O (NMO · H₂O; 310 mg, 2.29 mmol) in acetone/H₂O 3:2 (20 ml) was treated with OsO₄ (ca. 20 mg, ca. 0.08 mmol) and stirred at 25° for 2 h. Normal workup (AcOEt/Na₂SO₃ soln.; Na₂SO₄) and FC (hexane/AcOEt 1:1) gave (±)-**15** (565 mg, 81%). *R*_f (hexane/AcOEt 1:1) 0.51. IR (CH₂Cl₂): 3500s, 3366s, 2933m, 2855m, 1678s, 1461m, 1094s, 1011m, 839s. ¹H-NMR (CDCl₃): see Table 2; additionally, 7.35–7.27 (*m*, 5 H); 5.77, 4.78 (2*d*, *J* = 7.8, OCH₂O); 5.35 (*d*, *J* = 2.5, HO–C(5)); 5.12, 4.13 (2*d*, *J* = 14.6, PhCH₂); 3.97 (irrad. at 1.66 → NOE of ca. 2%, H–C(5)); 3.95 (irrad. at 3.46 → *t*, *J* ≈ 8.4, irrad. at 1.66 → NOE of ca. 2%, H–C(6)); 3.87 (*dd*, *J* = 9.7, 4.0, CH_a–C(8)); 3.73 (br. *d*, *J* ≈ 5.5, HO–C(6)); 3.52 (*s*, MeO); 3.46 (irrad. at 1.66 → NOE of 4.9%, H–C(1)); 3.27 (*t*, *J* = 9.5, CH_b–C(8)); 2.06 (*ddd*, *J* = 9.7, 5.6, 4.0, irrad. at 1.66 → NOE of 7.6%, H–C(8)); 1.66 (*ddd*, *J* = 14.0, 10.0, 2.2, irrad. at 3.46 → *dd*, *J* = 14.0, 10.0, H_{endo}–C(7)); 1.39 (*ddd*, *J* = 14.0, 5.9, 3.7, irrad. at 3.46 → *dd*, *J* = 14.0, 5.6, irrad. at 1.66 → NOE of 24.6%, H_{exo}–C(7)); 0.84 (*s*, *t*-Bu); – 0.02, – 0.03 (2*s*, Me₂Si). ¹³C-NMR (CDCl₃): 168.79 (*s*, C(3)); 137.53 (*s*); 128.91 (2*d*); 128.86 (2*d*); 127.84 (*d*); 93.58 (*t*, OCH₂O); 81.27 (*s*, C(4)); 71.43, 68.84 (2*d*, C(5), C(6)); 63.89 (*t*, CH₂–C(8)); 57.10 (*d*, C(1)); 55.88 (*q*, MeO), 49.70 (*t*, PhCH₂); 39.58 (*d*, C(8)); 28.09 (*t*, C(7)); 25.92 (*q*, Me₃CSi); 18.29 (*s*, Me₃CSi); – 5.51 (*q*, Me₂Si). CI-MS: 452 (6, [M + H]⁺), 420 (39), 362 (66), 334 (40), 91 (100), 75 (77), 49 (57). Anal. calc. for C₂₅H₃₇NO₆Si (451.63): C 61.17, H 8.26, N 3.10; found: C 61.17, H 8.06, N 3.14.

rac-(1*R*,4*R*,5*R*,6*R*,8*R*)-2-Benzyl-8-[[*tert*-butyl]dimethylsilyloxy]methyl]-5-hydroxy-4-(methoxymethoxy)-3-oxo-2-azabicyclo[2.2.2]oct-6-yl Trifluoromethanesulfonate ((±)-**16**) and Its Transformation to (±)-**14**. A cooled (–10°) soln. of (±)-**15** (1.01 g, 2.24 mmol) in pyridine (13 ml) was treated with Tf₂O (0.48 ml, 2.92 mmol) and stirred for 30 min. Normal workup (AcOEt/NaHCO₃ soln.; Na₂SO₄) and FC (cyclohexane/AcOEt 7:3) gave (±)-**16** (1.05 g, 81%). Pale white foam. *R*_f (hexane/AcOEt 7:3) 0.35. ¹H-NMR (CDCl₃): see Table 2; additionally, 7.33–7.29 (*m*, 5 H); 5.78, 4.75 (2*d*, *J* = 8.0, OCH₂O); 5.21 (*d*, *J* = 3.7, HO–C(5)); 4.95, 4.25 (2*d*, *J* = 14.7, PhCH₂); 4.90 (irrad. at 3.63 → *d*, *J* = 7.9, H–C(6)); 4.08 (irrad. at 5.21 → *d*, *J* = 7.9, H–C(5)); 3.82 (*dd*, *J* = 9.8, 3.9, CH_a–C(8)); 3.55 (*s*, MeO); 3.40 (*dd*, *J* = 9.7, 8.4, CH_b–C(8)); 2.06–1.98 (*m*, H–C(8)); 1.72 (*ddd*, *J* = 14.3, 9.6, 2.2, irrad. at 3.63 → *dd*, *J* ≈ 14.3, 9.6, H_{endo}–C(7)); 1.64 (*ddd*, *J* = 14.3, 6.5, 4.0, irrad. at 3.63 → *dd*, *J* ≈ 14.3, 6.5, H_{exo}–C(7)); 0.85 (*s*, *t*-Bu); 0.03, 0.01 (2*s*, Me₂Si). ¹³C-NMR (CDCl₃): 168.05 (*s*, C(3)); 136.25 (*s*); 129.0 (2*d*); 128.96 (2*d*); 128.15 (*d*); 118.20 (*q*, *J*(C,F) ≈ 320, CF₃); 93.87 (*t*, OCH₂O); 82.88 (*d*, C(6)); 80.82 (*s*, C(4)); 72.19 (*d*, C(5)); 63.13 (*t*, CH₂–C(8)); 56.15 (*q*, MeO); 54.33 (*d*, C(1)); 49.17 (*t*, PhCH₂); 39.42 (*d*, C(8)); 27.66 (*t*, C(7)); 26.08 (*q*, Me₃CSi); 18.51 (*s*, Me₃CSi); – 5.26, – 5.38 (2*q*, Me₂Si).

A soln. of (±)-**16** (21 mg, 0.035 mmol) and NaN₃ (35 mg, 0.53 mmol) in DMF (1 ml) was stirred at 85° for 4 h. Normal workup (AcOEt/H₂O; Na₂SO₄) and FC (cyclohexane/AcOEt 8:2) gave (±)-**14** (13 mg, 78%).

rac-(1*R*,4*R*,5*R*,6*R*,8*R*)-2-Benzyl-8-[[*tert*-butyl]dimethylsilyloxy]methyl]-4-(methoxymethoxy)-3-oxo-6-[[trifluoromethyl]sulfonyloxy]-2-azabicyclo[2.2.2]oct-5-yl Acetate ((±)-**18**). A cooled (0°) soln. of (±)-**15** (210 mg, 0.47 mmol) in pyridine (3 ml) was treated with Tf₂O (0.17 ml, 0.61 mmol), stirred at 25° for 7 h, treated with Ac₂O (0.3 ml, 1.7 mmol), and stirred at 25° for 12 h. Normal workup (AcOEt/H₂O; Na₂SO₄) and FC (hexane/AcOEt 17:3) gave (±)-**18** (180 mg, 66%). *R*_f (hexane/AcOEt 7:3) 0.85. IR (CHCl₃): 3004*m*, 2968*m*, 1727*s*, 1687*s*, 1473*m*, 1418*s*, 1120*s*, 1009*m*. ¹H-NMR (CDCl₃): see Table 2; additionally, 7.36–7.26 (*m*, 5 H); 5.39, 4.81 (2*d*, *J* = 7.5, OCH₂O); 5.34 (irrad. at 4.98 → *s*, H–C(5)); 5.03, 4.23 (2*d*, *J* = 14.8, PhCH₂); 4.98 (irrad. at 3.71 → *d*, *J* = 8.1, H–C(6)); 3.83 (*dd*, *J* = 10.0, 3.7, CH_a–C(8)); 3.71 (irrad. at 4.98 → br. *s*, H–C(1)); 3.42 (*dd*, *J* = 10.0, 8.1, CH_b–C(8)); 3.36 (*s*, MeO); 2.26–2.14 (*m*, H–C(8)); 2.11 (*s*, Ac); 1.76 (*ddd*, *J* = 12.3, 10.0, 2.2, irrad. at 3.71 → *dd*, *J* = 12.1, 10.0, H_{endo}–C(7)); 1.65 (*ddd*, *J* = 10.3, 6.2, 4.1, irrad. at 3.71 → *dd*, *J* = 10.3, 6.2, H_{exo}–C(7)); 0.85 (*s*, *t*-Bu); 0.03, 0.00 (2*s*, Me₂Si). ¹³C-NMR (CDCl₃): 170.13, 167.88 (2*s*, 2 C=O); 136.16 (*s*); 129.07 (2*d*); 128.94 (2*d*); 128.33 (*d*); 118.45 (*q*, *J*(C,F) = 321, CF₃); 93.90 (*t*, OCH₂O); 80.76 (*s*, C(4)); 77.74, 71.37 (2*d*, C(5), C(6)); 62.92 (*t*, CH₂–C(8)); 55.98 (*q*, MeO); 54.86 (*d*, C(1)); 49.21 (*t*, PhCH₂); 38.88 (*d*, C(8)); 27.49 (*t*, C(7)); 25.89 (*q*, Me₃CSi); 20.32 (*s*, Me₃CSi); 18.32 (*q*, MeC=O); – 5.51, – 5.61 (2*q*, Me₂Si). ¹⁹F-NMR (282 MHz, CDCl₃): – 74.77 (*s*). HR-MALDI-MS: 648.1883 (100, [M + Na]⁺, [C₂₆H₃₈F₃NNaO₉SSi]⁺; calc. 648.1886), 594.1796 (94, [M – MeO]⁺, [C₂₅H₃₅F₃NO₈SSi]⁺; calc. 594.1805).

rac-(1*R*,4*R*,5*R*,6*S*,8*R*)-2-Benzyl-8-[[*tert*-butyl]dimethylsilyloxy]methyl]-4-(methoxymethoxy)-3-oxo-2-azabicyclo[2.2.2]octane-5,6-diyl Diacetate ((±)-**19**). A soln. of (±)-**18** (20 mg, 0.032 mmol) in DMF (0.5 ml) was treated with AcOCs (9 mg, 0.048 mmol) and stirred at 60° for 6 h. Normal workup (AcOEt/H₂O; Na₂SO₄) and FC (hexane/AcOEt 1:4) gave (±)-**19** (15 mg, 84%). Colourless solid. *R*_f (hexane/AcOEt 17:3) 0.33. M.p.

82–83°. ¹H-NMR (CDCl₃): see Table 2; additionally, 7.34–7.25 (*m*, 5 H); 5.19 (*d*, *J* = 7.4), 4.83 (*d*, *J* = 7.4, irradiat. at 5.05 → NOE of 0.6%) (OCH₂O); 5.05 (irradiat. at 4.52 → *s*, H–C(5)); 4.62, 4.56 (*2d*, *J* = 14.6, PhCH₂); 4.52 (irradiat. at 3.62 → *br. s*, irradiat. at 5.05 → NOE of 4.5%, H–C(6)); 3.86 (*dd*, *J* = 10.0, 4.0, irradiat. at 2.34 → *d*, *J* = 10.0, CH₂–C(8)); 3.62 (irradiat. at 4.52 → *br. s*, irradiat. at 4.52 → NOE of 6.8%, H–C(1)); 3.41 (*t*, *J* = 10.0, irradiat. at 2.34 → *d*, *J* = 9.8, CH_b–C(8)); 3.38 (*s*, MeO); 2.39–2.29 (*m*, irradiat. at 5.05 → NOE of 7.3%, H–C(8)); 2.05 (*s*, irradiat. at 4.52 → NOE of 1.3%, 2 Ac); 2.00 (*ddd*, *J* ≈ 14.0, 10.0, 2.2, irradiat. at 2.34 → *br. d*, *J* ≈ 12.5, irradiat. at 3.62 → *dd*, *J* = 14.0, 10.3, H_{endo}–C(7)); 1.60–1.49 (*m*, irradiat. at 2.34 → *br. d*, *J* ≈ 14.0, irradiat. at 3.62 → *br. dd*, *J* ≈ 14.0, 5.6, irradiat. at 4.52 → *ddd*, *J* ≈ 14.0, 6.0, 3.0, H_{exo}–C(7)); 0.86 (*s*, *t*-Bu); 0.03, 0.01 (2*s*, Me₂Si). HR-MALDI-MS: 558.2490 (94, [M + Na]⁺, [C₂₇H₄₁NNaO₈Si]⁺; calc. 558.2499), 504.2408 (100, [M – MeO]⁺, [C₂₆H₃₈NO₇Si]⁺; calc. 504.2418), 416.1887 (27).

(1*R*,2*S*,5*R*)-5-Methyl-2-(1-methyl-1-phenylethyl)cyclohexyl (1*R*,4*R*,5*S*,7*R*,8*R*)-2-Benzyl-7,8-dihydroxy-4-(methoxymethoxy)-3-oxo-2-azabicyclo[2.2.2]octane-5-carboxylate (**21** [12]). A yellow soln. of **20** (3.00 g, 5.64 mmol) and NMO · H₂O (1.18 g, 8.46 mmol) in THF/acetone/H₂O 3 : 1 : 2 (60 ml) was treated with a 2.5% soln. of OsO₄ in *t*-BuOH (0.71 ml, 0.056 mmol) and stirred at 24° for 6 h. Normal workup (AcOEt/Na₂SO₃ soln.; brine; Na₂SO₄) and recrystallisation of the crude product from *i*-PrOH/hexane gave **21** (2.44 g, 77%). Colourless crystals. M.p. 162° (Et₂O/hexane; [12]; 162°).

(1*R*,2*S*,5*R*)-5-Methyl-2-(1-methyl-1-phenylethyl)cyclohexyl (1*R*,4*R*,5*S*,7*R*,8*R*)-8-Acetoxy-2-benzyl-4-(methoxymethoxy)-3-oxo-7-(trifluoromethyl)sulfonyloxy]-2-azabicyclo[2.2.2]octane-5-carboxylate (**22**) and (1*R*,2*S*,5*R*)-5-Methyl-2-(1-methyl-1-phenylethyl)cyclohexyl (1*R*,4*R*,5*S*,7*R*,8*R*)-2-Benzyl-4-(methoxymethoxy)-3-oxo-7,8-bis(trifluoromethyl)sulfonyloxy]-2-azabicyclo[2.2.2]octane-5-carboxylate (**23**). A soln. of **21** [12] (3.00 g, 5.30 mmol) in CH₂Cl₂/pyridine 1 : 1 (60 ml) was treated dropwise with Tf₂O (0.96 ml, 5.82 mmol) at 24°, stirred for 12 h, treated with Ac₂O (1.0 ml, 10.6 mmol), and stirred at 25° for 4 h. Normal workup (AcOEt/NaHCO₃ soln.; Na₂SO₄) and FC (cyclohexane/AcOEt 6 : 1) gave **23** (0.45 g, 11%) and **22** (3.16 g, 81%) as colourless foams.

Data of **22**: R_f (cyclohexane/AcOEt 3 : 2) 0.64. M.p. 86–90° (Et₂O/hexane). [α]_D²⁵ = –59.3 (*c* = 1.0, CHCl₃). IR (CHCl₃): 2966*m*, 2928*m*, 2852*w*, 1766*m*, 1728*s*, 1701*s*, 1600*w*, 1496*w*, 1454*w*, 1422*m*, 1371*m*, 1322*m*, 1143*s*, 1079*m*, 999*m*, 961*m*, 897*m*, 842*w*. ¹H-NMR (CDCl₃): see Table 3; additionally, 7.43–7.16 (*m*, 9 H); 7.15–7.08 (*m*, 1 H); 5.20, 4.13 (2*d*, *J* ≈ 14.8, PhCH₂); 5.07, 4.64 (2*d*, *J* = 7.4, OCH₂O); 4.81 (*td*, *J* = 10.8, 4.4, H–C(1′)); 4.80 (irradiat. at 3.62 → *d*, *J* = 7.5, H–C(7)); 3.30 (*s*, MeO); 2.08 (*td*, *J* ≈ 12.5, 3.7, irradiat. at 4.81 → *m*, H–C(2′)); 2.08 (*s*, Ac); 1.96 (*br. d*, *J* ≈ 10.9, irradiat. at 4.81 → *d*, *J* = 10.6, H_{eq}–C(6′)); 1.91 (*dd*, *J* = 10.3, 5.9, H–C(5)); 1.86 (*dq*, *J* ≈ 13.4, 3.7, H_{eq}–C(3′)); 1.72 (*ddd*, *J* = 14.0, 5.9, 3.7, irradiat. at 3.62 → *dd*, *J* = 14.3, 5.9, H_{exo}–C(6)); 1.72–1.63 (*m*, H_{eq}–C(4′)); 1.46 (*ddd*, *J* = 14.0, 10.3, 2.2, irradiat. at 3.62 → *dd*, *J* = 14.0, 10.0, H_{endo}–C(6)); 1.52–1.38 (*m*, H–C(5′)); 1.24, 1.16 (2*s*, Me₂PhC); 1.17 (*qd*, *J* ≈ 12.6, 3.4, H_{ax}–C(3′)); 0.91 (*qd*, *J* ≈ 11.6, 3.6, H_{ax}–C(4′)); 0.90 (*q*, *J* ≈ 12.2, irradiat. at 4.81 → *m*, H_{ax}–C(6′)); 0.88 (*d*, *J* = 6.5, Me–C(5′)). ¹³C-NMR (CDCl₃): 170.31, 169.54, 165.77 (3*s*, 3 C=O); 152.16, 135.15 (2*s*); 128.90 (2*d*); 128.63 (2*d*); 127.94 (*d*); 127.60 (2*d*); 125.08 (2*d*); 124.74 (*d*); 116.97 (*q*, *J*(C,F) = 316, CF₃); 93.38 (*t*, OCH₂O); 79.90 (*d*, C(8)); 77.87 (*s*, C(4)); 75.68 (*d*, C(1′)); 70.39 (*d*, C(7)); 56.05 (*q*, MeO); 53.40 (*d*, C(1)); 50.27 (*d*, C(2′)); 49.09 (*t*, PhCH₂); 41.55 (*t*, C(6′)); 41.39 (*d*, C(5)); 39.44 (*s*, Me₂PhC); 34.55 (*t*, C(4′)); 31.26 (*d*, C(5′)); 29.78, 22.75 (2*q*, Me₂PhC); 27.60 (*t*, C(6)); 26.22 (*t*, C(3′)); 21.83 (*q*, Me–C(5′)); 20.29 (*q*, MeC=O). ¹⁹F-NMR (282 MHz, CDCl₃): –74.80 (*s*). HR-MALDI-MS: 762.2536 (52, [M + Na]⁺, [C₃₆H₄₄F₃NNaO₁₀S]⁺; calc. 762.2536), 548.0892 (100, [M – (8-phenylmenthyl) + H + Na]⁺, [C₂₆H₂₂F₃NNaO₁₀S]⁺; calc. 548.0814), 494.0814 (80, [M – (8-phenylmenthyl) – MeO + H]⁺, [C₁₉H₁₉F₃NO₉S]⁺; calc. 494.0733). Anal. calc. for C₃₆H₄₄F₃NO₁₀S (739.80): C 58.45, H 5.99, N 1.89, S 4.33; found: C 58.50, H 6.10, N 1.90, S 4.33.

Data of **23**: R_f (cyclohexane/AcOEt 3 : 2) 0.76. M.p. 86–90°. [α]_D²⁵ = 10.0 (*c* = 1.0, CHCl₃). IR (CHCl₃): 2960*w*, 2928*w*, 1731*s*, 1709*m*, 1600*w*, 1496*w*, 1455*m*, 1428*s*, 1369*w*, 1324*m*, 1139*s*, 1092*m*, 1000*m*, 948*m*, 923*m*, 888*m*, 859*m*. ¹H-NMR (CDCl₃): see Table 3; additionally, 7.42–7.28 (*m*, 5 H); 7.24–7.16 (*m*, 4 H); 7.12 (*ddt*, *J* ≈ 8.1, 5.9, 2.6, 1 H); 5.42, 4.98 (2*d*, *J* = 14.8, PhCH₂); 5.41, 4.86 (2*d*, *J* ≈ 7.6, OCH₂O); 4.80 (*td*, *J* = 10.6, 4.4, H–C(1′)); 3.37 (*s*, MeO); 2.32 (*dd*, *J* = 10.6, 5.3, H–C(5)); 2.18 (*br. d*, *J* ≈ 11.5, H_{eq}–C(6′)); 2.06 (*td*, *J* = 10.6, 3.5, H–C(2′)); 1.67–1.50 (*m*, H_{eq}–C(3′), H_{eq}–C(4′)); 1.58 (*ddd*, *J* = 14.3, 10.7, 1.9, H_{endo}–C(6)); 1.50–1.31 (*m*, H–C(5′)); 1.33 (*dt*, *J* ≈ 14.3, 5.0, H_{exo}–C(6)); 1.19, 1.12 (2*s*, Me₂PhC); 1.05 (*qd*, *J* ≈ 12.5, 2.5, H_{ax}–C(3′)); 0.95 (*q*, *J* ≈ 11.2, H_{ax}–C(6′)); 0.87 (*d*, *J* = 6.5, Me–C(5′)); 0.85 (*qd*, *J* ≈ 11.6, 3.6, H_{ax}–C(4′)). ¹³C-NMR (CDCl₃): 169.91, 164.44 (2*s*, 2 C=O); 150.81, 135.15 (2*s*); 128.49 (2*d*); 128.45 (2*d*); 127.81 (*d*); 127.48 (2*d*); 124.94 (2*d*); 124.74 (*d*); 118.10, 117.94 (2*q*, *J*(C,F) ≈ 312, 2 CF₃); 93.51 (*t*, OCH₂O); 80.48, 77.79, 76.91 (3*d*, C(7), C(8), C(1′)); 76.31 (*s*, C(4)); 55.59 (*q*, MeO); 52.52 (*d*, C(1)); 49.32 (*d*, C(2′)); 48.70 (*t*, PhCH₂); 42.12 (*d*, C(5)); 40.63 (*t*, C(6′)); 39.59 (*s*, Me₂PhC); 34.15 (*t*, C(4′)); 31.02 (*d*, C(5′)); 27.67 (*t*, C(6)); 26.75 (*t*, C(3′)); 26.64, 26.56 (2*q*, Me₂PhC); 21.50 (*q*, Me–C(5′)). ¹⁹F-NMR (282 MHz, CDCl₃): –73.97, –73.49 (2*s*). HR-MALDI-MS:

Table 3. Selected $^1\text{H-NMR}$ Chemical Shifts [ppm] and Coupling Constants [Hz] of the Enantiomerically Pure Isoquinuclidines **3**, **4**, and **22–28**.

	3	4	22	23	24	25	26	27	28
H–C(1)	4.02	3.78	3.62	3.74	3.63	2.82	2.78	3.51	3.41
H _{endo} –C(7)	–	–	4.80	4.87	4.47	–	–	3.45	1.99
H _{exo} –C(7)	2.71	2.69	–	–	–	5.02	5.09	–	1.41
C(8)	3.48	3.39	5.20	4.99	4.83	4.96	4.80	5.13	5.00
$J(1,6_{endo})$	1.9	2.8	2.2	1.9	2.5	2.5	2.6	2.5	2.2
$J(1,6_{exo})$	^{a)}	3.1	3.7	4.4	4.0	2.8	3.1	4.0	^{b)}
$J(1,7_{endo})$	–	–	1.7	1.9	–	–	–	–	1.8
$J(1,7_{exo})$	^{a)}	3.6	–	–	4.4	4.0	3.9	2.1	3.8
$J(3,8)$	^{a)}	1.3	–	–	–	–	1.6	–	–
$J(6_{exo},7_{exo})$	^{a)}	1.2	–	–	1.9	2.5	1.9	2.2	2.2
$J(7_{endo},7_{exo})$	–	–	–	–	–	–	–	–	15.6
$J(7_{endo},8)$	–	–	7.8	7.8	–	–	–	–	8.1
$J(7_{exo},8)$	^{a)}	1.5	–	–	0.6	1.9	2.2	2.2	1.2

^{a)} Broad signals; $J(1,6_{exo}) < 3.5$, $J(1,7_{exo}) \approx 3.5$, $J(3,8) \approx J(6_{exo},7_{exo}) \approx J(7,8) < 2.5$ Hz deduced from the width of the signals at the half of their height. ^{b)} 3.41 (br. s, H–C(1)); 1.83–1.76 (*m*, H_{exo}–C(6)).

852.1914 (88, $[M + \text{Na}]^+$, $[\text{C}_{35}\text{H}_{41}\text{F}_6\text{NNaO}_{11}\text{S}_2]^+$; calc. 852.1923), 638.0170 (100, $[M - (8\text{-phenylmethyl}) + \text{H} + \text{Na}]^+$, $[\text{C}_{19}\text{H}_{19}\text{F}_6\text{NNaO}_{11}\text{S}_2]^+$; calc. 638.0201), 584.0105 (94, $[M - (8\text{-phenylmethyl}) - \text{MeO} + \text{H}]^+$, $[\text{C}_{18}\text{H}_{16}\text{F}_6\text{NO}_{10}\text{S}_2]^+$; calc. 584.0120).

(*1R,2S,5R*)-5-Methyl-2-(1-methyl-1-phenylethyl)cyclohexyl (*1R,4R,5S,7S,8R*)-7,8-Diacetoxy-2-benzyl-4-(methoxymethoxy)-3-oxo-2-azabicyclo[2.2.2]octane-5-carboxylate (**24**). A soln. of **22** (2.50 g, 3.38 mmol) and AcOCs (973 mg, 5.07 mmol) in DMF (50 ml) was stirred at 80° for 2.5 h. Normal workup (AcOEt/H₂O; Na₂SO₄) and FC (cyclohexane/AcOEt 3:1) gave **24** (2.05 g, > 98%) as a colourless solid. Recrystallisation from CH₂Cl₂/hexane (1:3) gave **24** (1.92 g, 94%). Colourless needles. *R*_f (cyclohexane/AcOEt 3:2) 0.39. *M*.p. 174.5° (CH₂Cl₂/hexane). $[\alpha]_D^{25} = 19.6$ (*c* = 1.0, CHCl₃). IR (CHCl₃): 2957*m*, 2927*m*, 1741*s* (br.), 1697*s*, 1600*w*, 1496*m*, 1454*m*, 1373*s*, 1330*m*, 1299*m*, 1159*s*, 1129*m*, 1090*m*, 1056*s*, 1038*s*, 978*m*, 959*m*, 916*m*, 850*w*. $^1\text{H-NMR}$ (CDCl₃): see Table 3; additionally, 7.40–7.18 (*m*, 9 H); 7.14–7.04 (*m*, 1 H); 4.99, 4.65 (2*d*, *J* = 7.5, OCH₂O); 4.83 (irrad. at 3.63 → *d*, *J* = 0.6, H–C(8)); 4.78 (*td*, *J* = 10.6, 4.3, H–C(1′)); 4.67, 4.55 (2*d*, *J* = 14.6, PhCH₂); 4.47 (irrad. at 3.63 → br. s, H–C(7)); 3.29 (*s*, MeO); 2.10, 2.09 (2*s*, 2 Ac); 2.06 (*dd*, *J* ≈ 10.0, 6.5, H–C(5)); 2.14–2.04 (*m*, irrad. at 4.78 → change, H_{eq}–C(6′)); 2.00 (br. *d*, *J* ≈ 13.7, irrad. at 4.81 → br. s, H–C(2′)); 1.84–1.73 (*m*, H_{eq}–C(3′)); 1.78 (*ddd*, *J* = 13.7, 10.0, 2.5, H_{endo}–C(6)); 1.70–1.59 (*m*, H_{eq}–C(4′)); 1.66 (*m*, irrad. at 3.63 → *ddd*, *J* ≈ 13.4, 6.7, 1.9, H_{exo}–C(6)); 1.53–1.35 (*m*, H–C(5′)); 1.24, 1.16 (2*s*, Me₂PhC); 1.12 (*qd*, *J* ≈ 12.8, 3.1, H_{ax}–C(3′)); 0.98–0.80 (*m*, H_{ax}–C(4′)); 0.89 (*q*, *J* ≈ 11.7, irrad. at 4.78 → *m*, H_{ax}–C(6′)); 0.87 (*d*, *J* = 6.5, Me–C(5′)). $^{13}\text{C-NMR}$ (CDCl₃): 170.96, 169.35, 168.84, 166.62 (4*s*, 4 C=O); 152.25, 135.97 (2*s*); 128.63 (2*d*); 128.52 (2*d*); 127.78 (*d*); 127.56 (2*d*); 125.10 (2*d*); 124.55 (*d*); 93.10 (*t*, OCH₂O); 78.17, 75.41, 74.99 (3*d*, C(7), C(8), C(1′)); 77.00 (*s*, C(4)); 55.92 (*q*, MeO); 51.06 (*d*, C(1)); 50.30 (*d*, C(2′)); 48.42 (*t*, PhCH₂); 42.45 (*d*, C(5)); 41.51 (*t*, C(6′)); 39.48 (*s*, Me₂PhC); 34.61 (*t*, C(4′)); 31.26 (*d*, C(5′)); 29.44, 23.13 (2*q*, Me₂PhC); 26.67 (*t*, C(6)); 26.29 (*t*, C(3′)); 21.83 (*q*, Me–C(5′)); 20.90, 20.84 (2*q*, 2 MeC=O). ESI-MS: 1321 (8, $[2M + \text{Na}]^+$), 672 (100, $[M + \text{Na}]^+$), 650 (6, $[M + \text{H}]^+$). Anal. calc. for C₃₇H₄₇NO₉ (649.77): C 68.39, H 7.29, N 2.16; found: C 68.24, H 7.30, N 2.30.

X-Ray Crystal-Structure Analysis of 24. The crystals were obtained by azeotropic recrystallisation in CH₂Cl₂/hexane. Orthorhombic, *P*₂₁*P*₂₁*P*₂₁, *a* = 9.575(2) Å, *b* = 13.034(2) Å, *c* = 28.313(3) Å, *V* = 3533.5(10) Å³, *Z* = 4, *D*_{calc} = 1.221 Mg/m³. From a crystal of size 0.40 × 0.30 × 0.30 mm, 3644 reflexions were measured on an Enraf Nonius CAD-4 diffractometer (graphite monochromator, CuK_α radiation, λ = 1.54184 Å) at 293(2) K, by Dr. B. Schweizer (ETH-Zürich). *R* = 0.0355, *R*_w = 0.1094. The structure was solved by direct methods with SIR97 [25]. The non-H-atoms were refined anisotropically with SHELXL-97 [22]. The H-atoms were calculated and included in the structure factor calculation. Drawings of the molecule were done with ORTEP [23].

(*1R,4R,5R,7S,8R*)-7,8-Diacetoxy-2-benzyl-4-(methoxymethoxy)-2-azabicyclo[2.2.2]octane-5-methyl Acetate (**25**) and (*1R,2S,5R*)-5-Methyl-2-(1-methyl-1-phenylethyl)cyclohexyl Acetate. A mixture of **24** (250 mg,

0.385 mmol) and LiAlH_4 (160 mg, 3.85 mmol) in THF (12.5 ml) was heated to reflux for 5 h, cooled to 0° , treated dropwise with MeOH (3 ml), and evaporated. The residue was suspended in pyridine/Ac₂O 2:1 (4.5 ml), treated with DMAP (5 mg, 0.04 mmol), and stirred at 24° for 12 h. The mixture was filtered through *Celite*, and the residue was washed with AcOEt (3×5 ml). Normal workup ($\text{CH}_2\text{Cl}_2/\text{NaHCO}_3$ soln.; brine; Na_2SO_4) and FC (cyclohexane/AcOEt 5:1 \rightarrow 3:1) gave (1*R*,2*S*,5*R*)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyl acetate (97 mg, 92%) as a colourless oil and **25** (152 mg, 88%) as a colourless solid.

Data of 25: R_f (cyclohexane/AcOEt 3:2) 0.41. $[\alpha]_D^{25} = 21.5$ ($c = 1.2$, CHCl_3). IR (CHCl_3): 2958w, 2826w, 1736s, 1494w, 1453w, 1370m, 1151w, 1121w, 1080w, 1030m, 921w. $^1\text{H-NMR}$ (CDCl_3): see *Table 3*; additionally, 7.34–7.21 (*m*, 5 H); 4.70, 4.60 (*2d*, $J = 7.2$, OCH_2O); 4.38 (*dd*, $J = 10.9$, 4.5, $\text{CH}_a\text{-C}(5)$); 4.17 (*dd*, $J = 10.9$, 9.0, $\text{CH}_b\text{-C}(5)$); 3.80, 3.75 (*2d*, $J = 13.6$, PhCH_2); 3.30 (*s*, MeO); 2.94 (*dd*, $J = 10.0$, 1.6, $\text{H}_a\text{-C}(3)$); 2.65 (*dd*, $J = 10.0$, 1.2, $\text{H}_b\text{-C}(3)$); 2.36–2.20 (*m*, $\text{H-C}(5)$); 2.13, 2.08, 2.05 (*3s*, 3 Ac); 1.94 (*ddd*, $J = 13.7$, 10.3, 2.5, $\text{H}_{endo}\text{-C}(6)$); 1.67 (*dddd*, $J = 14.0$, 5.6, 2.8, 2.5, $\text{H}_{exo}\text{-C}(6)$). $^{13}\text{C-NMR}$ (CDCl_3): 170.79, 169.65, 169.38 (*3s*, 3 C=O); 138.11 (*s*); 128.11 (*2d*); 128.05 (*2d*); 126.91 (*d*); 91.24 (*t*, OCH_2O); 76.27, 75.16 (*2d*, C(7), C(8)); 75.54 (*s*, C(4)); 64.97 (*t*, $\text{CH}_2\text{-C}(5)$); 59.13 (*t*, PhCH_2); 55.87 (*q*, MeO); 51.35 (*d*, C(1)); 48.58 (*d*, C(3)); 34.91 (*d*, C(5)); 24.34 (*t*, C(6)); 21.04, 21.01, 20.97 (*3q*, 3 MeC=O). HR-MALDI-MS: 450.2118 (83, $[M + \text{H}]^+$, $[\text{C}_{23}\text{H}_{32}\text{NO}_8]^+$; calc. 450.2128), 448.1970 (26, $[M - \text{H}]^+$, $[\text{C}_{23}\text{H}_{30}\text{NO}_8]^+$; calc. 448.1971), 390.1914 (100, $[M - \text{OAc}]^+$, $[\text{C}_{21}\text{H}_{28}\text{NO}_6]^+$; calc. 390.1917). Anal. calc. for $\text{C}_{23}\text{H}_{31}\text{NO}_8$ (449.50): C 61.46, H 6.95, N 3.12; found: C 61.47, H 7.03, N 2.95.

*Data of (1*R*,2*S*,5*R*)-5-Methyl-2-(1-methyl-1-phenylethyl)cyclohexyl Acetate*: see [26].

(1*R*,4*R*,5*R*,7*S*,8*R*)-7,8-Diacetoxy-2-benzyl-4-hydroxy-2-azabicyclo[2.2.2]octane-5-methyl Acetate (**26**). A cooled (0°) mixture of **25** (1.20 g, 2.67 mmol) and molecular sieves (4 Å; 0.60 g) in CH_2Cl_2 (60 ml) was treated dropwise with Me_3SiBr (0.71 ml, 5.38 mmol) and heated to reflux for 2 h. Normal workup (AcOEt/NaHCO₃ soln.; brine; Na_2SO_4) and FC (cyclohexane/AcOEt 2:1 \rightarrow 1:1) gave **26** (955 mg, 88%). Colourless oil. R_f (cyclohexane/AcOEt 1:2) 0.60. $[\alpha]_D^{25} = 19.4$ ($c = 1.0$, CHCl_3). IR (CHCl_3): 3577w, 2961w, 2837w, 1735s, 1494w, 1453w, 1369m, 1125w, 1079w, 1031m, 977w, 914w. $^1\text{H-NMR}$ (CDCl_3): see *Table 3*; additionally, 7.33–7.21 (*m*, 5 H); 4.41 (*dd*, $J = 11.2$, 6.2, $\text{CH}_a\text{-C}(5)$); 4.23 (*dd*, $J = 11.2$, 7.4, $\text{CH}_b\text{-C}(5)$); 3.81, 3.74 (*2d*, $J = 13.4$, PhCH_2); 2.74 (*dd*, $J = 10.0$, 1.6, $\text{H}_a\text{-C}(3)$); 2.65 (*dd*, $J = 10.0$, 1.6, $\text{H}_b\text{-C}(3)$); 2.31 (*s*, $\text{HO-C}(4)$); 2.15, 2.06, 2.04 (*3s*, 3 Ac); 2.15–2.06 (*m*, $\text{H-C}(5)$); 1.95 (*ddd*, $J = 13.4$, 10.6, 2.6, $\text{H}_{endo}\text{-C}(6)$); 1.58 (*dddd*, $J = 13.6$, 5.3, 3.1, 1.9, $\text{H}_{exo}\text{-C}(6)$). $^{13}\text{C-NMR}$ (CDCl_3): 171.37, 170.77, 169.40 (*3s*, 3 C=O); 138.22 (*s*); 128.23 (*4d*); 127.05 (*d*); 79.18, 74.34 (*2d*, C(7), C(8)); 71.75 (*t*, $\text{CH}_2\text{-C}(5)$); 65.08 (*s*, C(4)); 59.13 (*t*, PhCH_2); 51.40 (*d*, C(1)); 50.11 (*t*, C(3)); 37.49 (*d*, C(5)); 24.45 (*t*, C(6)); 21.13 (*q*, 3 MeCO). HR-MALDI-MS: 406.1856 (73, $[M + \text{H}]^+$, $[\text{C}_{21}\text{H}_{28}\text{NO}_7]^+$; calc. 406.1866), 404.1701 (26, $[M - \text{H}]^+$, $[\text{C}_{21}\text{H}_{26}\text{NO}_7]^+$; calc. 404.1709), 346.1651 (100, $[M - \text{OAc}]^+$, $[\text{C}_{19}\text{H}_{24}\text{NO}_5]^+$; calc. 346.1654). Anal. calc. for $\text{C}_{21}\text{H}_{27}\text{NO}_7$ (405.45): C 62.21, H 6.71, N 3.45; found: C 61.94, H 6.91, N 3.23.

(1*R*,4*R*,5*R*,7*S*,8*R*)-2-Benzyl-5-(hydroxymethyl)-2-azabicyclo[2.2.2]octane-4,7,8-triol (**3**). A cooled (0°) soln. of **26** (50 mg, 0.123 mmol) in MeOH (20 ml) was saturated with NH_3 and stirred at 24° for 12 h. Ion-exchange chromatography (*Amberlite CG-120*, H^+ form, 0.1M aq. NH_3) gave **3** (34 mg, 99%). Colourless solid. R_f (AcOEt/MeOH 10:1) 0.30. $[\alpha]_D^{25} = 8.0$ ($c = 0.5$, EtOH). IR (0.5% in KBr): 3374s (br.), 2899m (br.), 1495w, 1452m, 1380m, 1150m, 1111m, 1038s, 975m, 926w, 803w. $^1\text{H-NMR}$ (300 MHz, D_2O): see *Table 3*; additionally, 7.44–7.33 (*m*, 5 H); 3.89 (*dd*, $J = 10.6$, 5.6, $\text{CH}_a\text{-C}(5)$); 3.83, 3.79 (*2d*, $J \approx 13.3$, PhCH_2); 3.70 (*dd*, $J = 10.9$, 7.8, $\text{CH}_b\text{-C}(5)$); 2.73 (br. *d*, $J = 10.3$, $\text{H}_a\text{-C}(3)$); 2.60 (br. *d*, $J = 10.6$, $\text{H}_b\text{-C}(3)$); 2.00 (*ddd*, $J = 13.1$, 11.2, 1.9, $\text{H}_{endo}\text{-C}(6)$); 1.87 (br. *tt*, $J = 10.9$, 6.0, $\text{H-C}(5)$); 1.63 (br. *d*, $J \approx 13.0$, $w_{1/2} = 12$, $\text{H}_{exo}\text{-C}(6)$). $^{13}\text{C-NMR}$ (75 MHz, D_2O): 139.73 (*s*); 131.87 (*2d*); 131.11 (*2d*); 130.14 (*d*); 82.21, 76.54 (*2d*, C(7), C(8)); 74.91 (*s*, C(4)); 64.41 (*t*, $\text{CH}_2\text{-C}(5)$); 60.96 (*t*, PhCH_2); 55.50 (*d*, C(1)); 50.77 (*t*, C(3)); 40.52 (*d*, C(5)); 25.11 (*t*, C(6)). HR-MALDI-MS: 302.1360 (3, $[M + \text{Na}]^+$, $[\text{C}_{15}\text{H}_{21}\text{NNaO}_4]^+$; calc. 302.1368), 280.1541 (100, $[M + \text{H}]^+$, $[\text{C}_{15}\text{H}_{22}\text{NO}_4]^+$; calc. 280.1549). Anal. calc. for $\text{C}_{15}\text{H}_{21}\text{NO}_4 \cdot \frac{1}{2} \text{H}_2\text{O}$ (288.34): C 62.48, H 7.69, N 4.86; found: C 62.56, H 7.60, N 4.89. $pK_{\text{HA}} = 6.6$.

(1*R*,4*R*,5*R*,7*S*,8*R*)-5-(Hydroxymethyl)-2-azabicyclo[2.2.2]octane-4,7,8-triol (**4**). A mixture of **3** (31 mg, 0.11 mmol), 20% Pd(OH)–C (6 mg), MeOH (3 ml), and conc. aq. HCl (0.15 ml) was stirred at 24° for 12 h under H_2 (6 bar). The mixture was filtered through *Celite*, and the residue was washed with MeOH (3×5 ml). The filtrate was diluted with toluene (5 ml) and evaporated. Ion-exchange chromatography (*Amberlite CG-120*, H^+ form, 0.1M aq. NH_3) gave **4** (21 mg, 95%). Colourless solid. R_f (MeOH/25% aq. NH_3 5:2) 0.49. $[\alpha]_D^{25} = +61.1$ ($c = 0.4$, H_2O). IR (0.5% in KBr): 3391s (br.), 2935m, 2890m, 1636w, 1530m, 1423m, 1384m, 1258w, 1172w, 1112m, 1045m, 1008m, 926w, 808w. $^1\text{H-NMR}$ (300 MHz, CD_3OD): see *Table 3*; additionally, 3.90 (*dd*, $J = 10.6$, 7.2, $\text{CH}_a\text{-C}(5)$); 3.62 (*dd*, $J = 10.9$, 7.2, $\text{CH}_b\text{-C}(5)$); 2.89 (*dd*, $J = 10.9$, 1.3, $\text{H}_a\text{-C}(3)$); 2.70 (*dd*, $J \approx 10.9$, 1.2, $\text{H}_b\text{-C}(3)$); 2.16 (*ddd*, $J = 14.0$, 11.2, 2.8, $\text{H}_{endo}\text{-C}(6)$); 1.84 (*dqd*, $J \approx 11.2$, 5.9, 1.3, $\text{H-C}(5)$); 1.38 (*dddd*, $J = 14.0$, 4.7, 3.1, 1.2, $\text{H}_{exo}\text{-C}(6)$). $^{13}\text{C-NMR}$ (75 MHz, CD_3OD): 81.59, 72.98 (*2d*, C(7), C(8)); 78.41 (*s*, C(4)); 63.84

(*t*, CH₂–C(5)); 50.29 (*t*, C(3)); 40.65, 39.95 (2*d*, C(1), C(5)); 26.19 (*t*, C(6)). ESI-MS: 401 (18, [2*M* + Na]⁺), 360 (43, [2*M* – H₂O]⁺), 250 (18), 212 (30, [M + Na]⁺), 190 (38, [M + H]⁺), 181 (43), 149 (100). Anal. calc. for C₃₅H₄₅NO₄ · ¼ H₂O (202.72): C 47.40, H 8.20, N 6.91; found: C 47.26, H 8.36, N 6.75. p*K*_{HA} = 7.80.

(1*R*,2*S*,5*R*)-5-Methyl-2-(1-methyl-1-phenylethyl)cyclohexyl (1*R*,4*R*,5*R*,7*S*,8*S*)-8-Acetoxy-2-benzyl-4-(methoxymethoxy)-7-iodo-3-oxo-2-azabicyclo[2.2.2]octane-5-carboxylate (**27**). A soln. of **22** (348 mg, 0.47 mmol) and Bu₃Ni (1.74 g, 4.70 mmol) in DMF (5 ml) was stirred at 80° for 6 h. Normal workup (Et₂O/10% aq. Na₂S₂O₃ soln.; Na₂SO₄) and FC (cyclohexane/AcOEt 4:1) gave **27** (322 mg, 95%). Colourless foam. *R*_f (cyclohexane/AcOEt 2:3) 0.48. M.p. 56–60°. [*α*]_D²⁵ = 35.8 (*c* = 0.7, CHCl₃). IR (CHCl₃): 3011*m*, 2959*s*, 2928*s*, 1750*s*, 1728*s*, 1697*s*, 1610*w*, 1496*m*, 1453*m*, 1371*s*, 1324*m*, 1133*m*, 1077*m*, 1045*s*, 992*m*, 926*w*, 849*w*. ¹H-NMR (CDCl₃): see Table 3; additionally, 7.37–7.17 (*m*, 10 H); 5.13 (irrad. at 3.51 → *s*, H–C(8)); 5.01, 4.59 (2*d*, *J* = 7.8, OCH₂O); 4.78 (*td*, *J* = 10.6, 4.4, H–C(1′)); 4.72, 4.48 (2*d*, *J* = 14.9, PhCH₂); 3.45 (irrad. at 3.51 → br. *s*, H–C(7)); 3.28 (*s*, MeO); 2.47 (*ddd*, *J* = 13.4, 10.0, 2.5, H_{endo}–C(6)); 2.10 (*td*, *J* ≈ 11.5, 3.4, irrad. at 4.78 → *m*, H–C(2′)); 2.09 (*dd*, *J* = 10.6, 7.5, H–C(5)); 2.04 (*s*, Ac); 2.04–1.92 (*m*, irrad. at 3.51 → change, irrad. at 4.78 → change, H_{exo}–C(6), H_{eq}–C(6′)); 1.89 (*dq*, *J* ≈ 12.8, 3.3, H_{eq}–C(3′)); 1.70 (br. *d*, *J* ≈ 12.5, H_{eq}–C(4′)); 1.50–1.37 (*m*, H–C(5′)); 1.25, 1.18 (2*s*, Me₂PhC); 1.17 (*qd*, *J* ≈ 12.8, 3.4, H_{ax}–C(3′)); 0.93 (*qd*, *J* ≈ 12.5, 3.4, H_{ax}–C(4′)); 0.89 (*q*, *J* ≈ 12.0, irrad. at 4.78 → *t*, *J* ≈ 12.8, H_{ax}–C(6′)); 0.88 (*d*, *J* = 6.5, Me–C(5′)). ¹³C-NMR (CDCl₃): 170.89, 169.26, 166.15 (3*s*, 3 C=O); 152.15, 135.95 (2*s*); 128.63 (2*d*); 128.49 (2*d*); 127.93 (*d*); 127.73 (2*d*); 125.17 (2*d*); 124.98 (*d*); 93.18 (*t*, OCH₂O); 81.50 (*d*, C(8)); 77.91 (*s*, C(4)); 75.48 (*d*, C(1′)); 55.93 (*q*, MeO); 55.92 (*d*, C(1)); 50.39 (*d*, C(2′)); 48.08 (*t*, PhCH₂); 41.50 (*d*, C(5)); 41.50 (*t*, C(6′)); 39.37 (*s*, Me₂PhC); 34.64 (*t*, C(4′)); 31.28 (*d*, C(5′)); 29.95, 22.44, 21.85; 20.62 (4*q*, Me₂PhC, Me–C(5′), MeC=O); 29.63 (*t*, C(6)); 26.22 (*t*, C(3′)); 24.10 (*d*, C(7)). HR-MALDI-MS: 740.2059 (69, [M + Na]⁺, [C₃₅H₄₄INNaO₇]⁺; calc. 740.2060), 526.0340 (100, [M – 8-phenylmethyl + H + Na]⁺, [C₁₉H₂₂INNaO₇]⁺; calc. 526.0339). Anal. calc. for C₃₅H₄₄INO₇ (717.64): C 58.58, H 6.18, N 1.95; found: C 58.63, H 6.37, N 1.88.

(1*R*,2*S*,5*R*)-5-Methyl-2-(1-methyl-1-phenylethyl)cyclohexyl (1*R*,4*R*,5*S*,8*S*)-8-Acetoxy-2-benzyl-4-(methoxymethoxy)-3-oxo-2-azabicyclo[2.2.2]octane-5-carboxylate (**28**). A soln. of **27** (250 mg, 0.35 mmol) and 2,2′-azobis(2-methylpropionitrile) (AIBN; 0.62 mg, 0.004 mmol) in toluene (1 ml) was treated with Bu₃SnH (0.19 ml, 0.70 mmol), stirred at 43° for 2 h, and evaporated. FC (cyclohexane/AcOEt 3:1) gave **28** (185 mg, 90%). Colourless foam. *R*_f (cyclohexane/AcOEt 2:3) 0.32. M.p. 59–65°. [*α*]_D²⁵ = 9.2 (*c* = 0.8, CHCl₃). IR (CHCl₃): 3013*m*, 2959*m*, 2928*m*, 1728*s*, 1690*s*, 1601*w*, 1496*w*, 1456*m*, 1180*m*, 1159*m*, 1092*m*, 1051*m*, 1017*m*, 977*m*, 960*m*, 924*w*, 847*w*. ¹H-NMR (CDCl₃): see Table 3; additionally, 7.36–7.20 (*m*, 9 H); 7.11 (*dddd*, *J* = 9.0, 7.8, 4.3, 3.5, 1 H); 5.08, 4.68 (2*d*, *J* = 7.5, OCH₂O); 4.86, 4.42 (2*d*, *J* = 14.9, PhCH₂); 4.81 (*td*, *J* = 10.8, 4.4, H–C(1′)); 3.32 (*s*, MeO); 2.11–1.98 (*m*, H_{eq}–C(6′)); 2.07 (*td*, *J* ≈ 10.6, 3.5, irrad. at 1.79 → *t*, *J* ≈ 10.6, H–C(2′)); 2.05 (br. *dd*, *J* = 10.0, 6.2, H–C(5)); 2.02 (*s*, Ac); 1.99 (irrad. at 5.00 → br. *d*, *J* ≈ 15.6, irrad. at 3.41 → *dd*, *J* = 15.6, 8.7, H_{endo}–C(7)); 1.83–1.76 (*m*, irrad. at 3.41 → change, H_{exo}–C(6)); 1.79 (*dq*, *J* ≈ 12.8, 3.2, H_{eq}–C(3′)); 1.67 (br. *ddd*, *J* ≈ 12.8, 3.4, 2.8, H_{eq}–C(4′)); 1.52 (*ddd*, *J* = 12.8, 10.3, 2.2, irrad. at 3.41 → *dd*, *J* = 13.1, 10.4, irrad. at 2.05 → *dd*, 10.3, 2.2, H_{endo}–C(6)); 1.48–1.35 (*m*, H–C(5′)); 1.41 (irrad. at 5.00 → *dt*, *J* ≈ 15.6, 4.3, irrad. at 3.41 → *dt*, *J* ≈ 14.6, 2.2, irrad. at 2.05 → br. *dd*, *J* ≈ 15.6, 3.8, H_{exo}–C(7)); 1.25, 1.17 (2*s*, Me₂PhC); 1.14 (*q*, *J* ≈ 12.7, irrad. at 1.79 → *td*, *J* ≈ 12.1, 2.2, H_{ax}–C(3′)); 0.91 (*q*, *J* ≈ 12.1, H_{ax}–C(6′)); 0.90 (*qd*, *J* ≈ 12.8, 3.4, irrad. at 1.79 → *q*, *J* ≈ 12.6, H_{ax}–C(4′)); 0.88 (*d*, *J* = 7.8, Me–C(5′)). ¹³C-NMR (CDCl₃): 171.30, 169.84, 166.88 (3*s*, 3 C=O); 152.07, 136.56 (2*s*); 128.42 (2*d*); 128.33 (2*d*); 127.54 (*d*); 127.48 (2*d*); 125.06 (2*d*); 124.64 (*d*); 93.11 (*t*, OCH₂O); 78.61 (*s*, C(4)); 75.24 (*d*, C(1′)); 72.96 (*d*, C(8)); 55.78 (*q*, MeO); 50.25 (*d*, C(2′)); 48.95 (*d*, C(1)); 48.01 (*t*, PhCH₂); 42.53 (*d*, C(5)); 41.53 (*t*, C(6′)); 39.47 (*s*, Me₂PhC); 36.72 (*t*, C(7)); 34.57 (*t*, C(4′)); 32.20 (*t*, C(6)); 31.23 (*d*, C(5′)); 26.30 (*t*, C(3′)); 29.18, 23.36, 21.80; 21.07 (4*q*, Me₂PhC, Me–C(5′), MeC=O). HR-MALDI-MS: 614.3082 (100, [M + Na]⁺, [C₃₅H₄₅NNaO₇]⁺; calc. 614.3094), 400.1360 (98, [M – 8-phenylmethyl + H + Na]⁺, [C₁₉H₂₃NNaO₇]⁺; calc. 400.1372), 346.1286 (55). Anal. calc. for C₃₅H₄₅NO₇ (591.73): C 71.04, H 7.66, N 2.37; found: C 71.11, H 7.74, N 2.25.

(1*R*,4*R*,5*R*,8*R*)-8-Acetoxy-2-benzyl-4-(methoxymethoxy)-2-azabicyclo[2.2.2]octane-5-methyl Acetate (**29**) and (1*R*,2*S*,5*R*)-5-Methyl-2-(1-methyl-1-phenylethyl)cyclohexyl Acetate. A mixture of **28** (150 mg, 0.25 mmol) and LiAlH₄ (85 mg, 2.23 mmol) in THF (10 ml) was heated to reflux for 90 min, cooled to 0°, treated dropwise with MeOH (20 ml), and evaporated. The residue was suspended in pyridine/Ac₂O 2:1 (30 ml), treated with DMAP (2 mg, 0.03 mmol), and stirred at 25° for 12 h. The mixture was filtered through *Celite*, and the residue was washed with AcOEt (3 × 5 ml). Normal workup (CH₂Cl₂/NaHCO₃ soln.; brine; Na₂SO₄) and FC (cyclohexane/AcOEt 3:1 → 1:1) gave (1*R*,2*S*,5*R*)-5-methyl-2-(1-methyl-1-phenylethyl)cyclohexyl acetate (70 mg, > 98%) and **29** (90 mg, 91%) as colourless oils.

Data of **29**: *R*_f (cyclohexane/AcOEt/MeOH 1:1:0.1) 0.30. [*α*]_D²⁵ = 13.2 (*c* = 0.9, CHCl₃). IR (CHCl₃): 2951*m*, 2925*w*, 1731*s*, 1602*w*, 1494*w*, 1453*m*, 1369*s*, 1149*s*, 1126*m*, 1045*s*, 922*m*, 830*w*. ¹H-NMR (CDCl₃): 7.34–

7.20 (*m*, 5 H); 5.02 (*ddd*, $J = 9.3, 3.1, 1.8$, H–C(8)); 4.75, 4.61 (*2d*, $J = 7.5$, OCH₂O); 4.42 (*dd*, $J \approx 10.7, 4.6$, CH_a–C(5)); 4.21 (*dd*, $J = 10.6, 9.4$, CH_b–C(5)); 3.69 (*s*, PhCH₂); 3.32 (*s*, MeO); 2.98 (*dd*, $J = 9.9, 1.6$, H_a–C(3)); 2.73 (*dd*, $J = 10.0, 1.9$, H_b–C(3)); 2.65 (*quint.*, $J \approx 2.5$, H–C(1)); 2.22 (*br. tt.*, $J \approx 9.7, 4.3$, H–C(5)); 2.13 (*ddd*, $J \approx 14.9, 9.7, 2.0$, H_{endo}–C(7)); 2.12, 2.08 (*2s, 2 Ac*); 1.89 (*dq*, $J \approx 14.3, 3.0$, H_{exo}–C(7)); 1.81 (*ddd*, $J \approx 13.4, 4.8, 2.8$, H_{exo}–C(6)); 1.71 (*ddd*, $J \approx 13.4, 10.6, 2.5$, H_{endo}–C(6)). ¹³C-NMR (CDCl₃): 171.01, 170.21 (*2s, 2 C=O*); 138.84 (*s*); 128.19 (*2d*); 128.13 (*2d*); 126.81 (*d*); 91.21 (*t*, OCH₂O); 75.14 (*s*, C(4)); 71.67 (*d*, C(8)); 65.57 (*t*, CH₂–C(5)); 59.49 (*t*, C(3)); 55.92 (*q*, MeO); 49.70 (*t*, PhCH₂); 48.46 (*d*, C(1)); 35.11 (*d*, C(5)); 33.48 (*t*, C(7)); 29.80 (*t*, C(6)); 21.39, 21.09 (*2q, 2 MeC=O*). HR-MALDI-MS: 414.1883 ([*M* + Na]⁺, [C₂₁H₃₀NNaO₆]⁺; calc. 414.1893), 392.2064 ([*M* + H]⁺, [C₂₁H₃₀NO₆]⁺; calc. 392.2073).

Data of (1*R*,2*S*,5*R*)-5-Methyl-2-(1-methyl-1-phenylethyl)cyclohexyl Acetate: see [26].

(1*R*,4*R*,5*R*,8*R*)-2-Benzyl-5-(hydroxymethyl)-2-azabicyclo[2.2.2]octane-4,8-diol (**30**). A soln. of **29** (38 mg, 0.10 mmol) and 37% aq. HCl (0.4 ml) in EtOH (4 ml) was stirred at 25° for 24 h and evaporated. Ion-exchange chromatography (Amberlite CG-120, 0.1M aq. NH₃) and FC (Macherey-Nagel Lichoprep-NH₂, AcOEt/MeOH 10:1) gave **30** (23 mg, 89%). Colourless solid. *R*_f (Macherey-Nagel Nanosil-NH₂; AcOEt/MeOH 6:1) 0.39. [α]_D²⁵ = –14.6 (*c* = 0.6, MeOH). IR (0.5% in KBr): 3400*m*, 3083*w*, 3028*w*, 2937*m*, 2892*m*, 2866*m*, 1956*w*, 1883*w*, 1810*w*, 1740*w*, 1602*w*, 1494*w*, 1453*w*, 1384*w*, 1351*w*, 1119*m*, 1031*m*, 990*w*, 956*w*, 916*w*, 817*w*. ¹H-NMR (CD₃OD): 7.35–7.26 (*m*, 4 H); 7.21 (*tt*, $J \approx 6.9, 1.9, 1$ H); 3.90 (*dd*, $J \approx 10.5, 6.5$, CH_a–C(5)); 3.73 (*dd*, $J \approx 10.5, 5.9$, CH_b–C(5)); 3.71, 3.67 (*2d*, $J \approx 13.4, 13.4$, PhCH₂); 3.72–3.67 (*m*, H–C(8)); 2.78 (*dd*, $J = 10.0, 1.5$, H_a–C(3)); 2.61 (*dd*, $J = 10.3, 2.2$, H_b–C(3)); 2.62–2.58 (*m*, H–C(1)); 2.00 (*ddd*, $J = 14.0, 9.3, 2.8$, H_{endo}–C(7)); 1.92 (*br. d*, $J \approx 13.7, 13.7$, H_{exo}–C(7)); 1.78 (*dqd*, $J \approx 13.7, 6.9, 1.0$, H–C(5)); 1.70–1.66 (*m*, 2 H–C(6)). ¹³C-NMR (CD₃OD): 139.80 (*s*); 129.76 (*2d*); 129.15 (*2d*); 127.95 (*d*); 73.32 (*s*, C(4)); 72.70 (*d*, C(8)); 63.93 (*t*, CH₂–C(5)); 60.52 (*t*, C(3)); 51.33 (*t*, PhCH₂); 50.55 (*d*, C(1)); 40.78 (*d*, C(5)); 34.08 (*t*, C(7)); 30.33 (*t*, C(6)). HR-MALDI-MS: 264.1596 ([*M* + H]⁺, [C₁₅H₂₁NO₃]⁺; calc. 264.1600). Anal. calc. for C₁₅H₂₁NO₃ · ½ H₂O (272.16): C 68.15, H 8.39, N 5.30; found: C 67.92, H 8.46, N 5.26.

(1*R*,4*R*,5*R*,8*R*)-5-(Hydroxymethyl)-2-azabicyclo[2.2.2]octane-4,8-diol (**31**). A mixture of **30** (23 mg, 0.09 mmol), 20% Pd(OH)–C (6 mg), MeOH (1 ml), and 37% aq. HCl (0.15 ml) was stirred at 24° for 16 h under H₂ (6 bar). The mixture was filtered through Celite, and the residue was washed with MeOH (3 × 5 ml). The filtrate was diluted with toluene (5 ml) and evaporated. Ion-exchange chromatography (Amberlite CG-120, 0.1M aq. NH₃) gave **31** (14 mg, 92%). Colourless solid. *R*_f (MeOH/25% aq. NH₃ soln. 3:1) 0.25. [α]_D²⁵ = +25.3 (*c* = 0.8, H₂O). IR (0.5% in KBr): 3293*m*, 3221*m*, 3045*m*, 2910*m*, 2776*m*, 2482*w*, 2138*w*, 1610*m*, 1480*w*, 1410*m*, 1337*m*, 1262*w*, 1155*m*, 1111*m*, 993*m*, 958*w*, 901*w*, 846*w*. ¹H-NMR (CD₃OD): 3.85 (*dd*, $J = 10.9, 4.7$, CH_a–C(5)); 3.79 (*br. d*, $J = 9.7$, H–C(8)); 3.76 (*dd*, $J = 10.9, 5.0$, CH_b–C(5)); 3.45 (*quint.*, $J \approx 2.8$, H–C(1)); 3.31 (*dd*, $J \approx 11.5, 1.5$, H_a–C(3)); 3.23 (*dd*, $J = 11.5, 1.6$, H_b–C(3)); 2.36 (*ddd*, $J = 14.9, 9.7, 1.9$, H_{endo}–C(7)); 1.99 (*br. d*, $J \approx 15.5, 15.5$, H_{exo}–C(7)); 1.95–1.85 (*m*, H–C(5), 2 H–C(6)). ¹³C-NMR (CD₃OD): 70.20 (*s*, C(4)); 70.05 (*d*, C(8)); 61.41 (*t*, CH₂–C(5)); 46.82 (*d*, C(1)); 39.82 (*t*, C(3)); 38.72 (*d*, C(5)); 35.20 (*t*, C(7)); 27.87 (*t*, C(6)). ESI-MS: 174 (45, [*M* + H]⁺), 163 (54), 153 (26), 145 (35), 103 (100), 97 (54).

Inhibition of Glycosidases. The *IC*₅₀ values were determined at a substrate concentration corresponding to the *K*_M of each enzyme. *IC*₅₀ Values were calculated by plotting the inhibitor concentration vs. the rate of hydrolysis. Determination of the inhibition constants *K*_i was performed at five different concentrations of the inhibitor bracketing the *IC*₅₀ value. *K*_i Values and α values were determined from the replot of the slopes and the replot of the 1/*v* axis intercepts of Lineweaver–Burk plots [27].

a) **Inhibition of C. saccharolyticum β -Glucosidase.** *K*_M = 0.80 mM ([28]; *K*_M = 0.50–0.78 mM; [18]; *K*_M = 0.51 mM). The inhibition studies were carried out at pH 6.8 (KH₂PO₄/K₂HPO₄ buffer, 0.06M) and 55°. The reaction was started by the addition of 4-nitrophenyl β -D-glucopyranoside after pre-incubating the enzyme in the presence of the inhibitor for 30 min at 55°. After 15 min, the enzyme reaction was quenched by the addition of borate buffer (pH 9.2, 0.2M). The rate of hydrolysis was determined by measuring the absorption at $\lambda = 405$ nm and subsequently subtracting the absorption of a blank probe (H₂O, buffer, substrate).

b) **Inhibition of Sweet Almonds β -Glucosidases.** *K*_M = 3.4 mM ([28]; *K*_M = 2.9–3.1 mM). The inhibition studies were carried out at pH 6.8 (KH₂PO₄/K₂HPO₄ buffer, 0.06M) and 37°. The reaction was started by the addition of 4-nitrophenyl β -D-glucopyranoside after pre-incubating the enzyme in the presence of the inhibitor for 30 min at 37°. The velocity of the substrate hydrolysis was determined by measuring the increase of absorption per min at $\lambda = 405$ nm during 10 min.

c) **Inhibition of T. reesei Cellulase Cel7A.** *K*_M = 0.39 mM ([29]; *K*_M = 0.406 mM). The inhibition studies were carried out at pH 5.7 (KH₂PO₄/Na₂HPO₄ buffer, 0.05M) and 30°. The reaction was started by the addition of 4-nitrophenyl β -lactopyranoside after pre-incubating the enzyme in the presence of the inhibitor for 30 min at 30°.

After 10 min, the enzyme reaction was quenched by the addition of borate buffer (pH 9.2, 0.2M). The rate of hydrolysis was determined by measuring the absorption at $\lambda = 405$ nm and subsequently subtracting the absorption of a blank probe (H₂O, buffer, substrate).

d) *Inhibition of N. takasagoensis Endoglucanase NtEgI*. $K_M = 8.7$ mg/ml (corresponding to ca. 0.032 mM of 'sodium carboxymethyl cellulose'). According to a personal communication from Prof. E. Meyer, Texas A&M University, the inhibition studies were carried out at pH 5.6 (AcONa buffer, 0.04M containing CaCl₂, 0.8 mM) and 37°. The reaction was started by the addition of a 2.3 (w/w) 'sodium carboxymethyl cellulose' soln. (average molecular weight ca. 250000 Da, degree of carboxymethyl substitution 0.7 (w/w)) after pre-incubating the enzyme in the presence of the inhibitor for 30 min at 37°. After 30 min, the mixture was added to a soln. of tetrazolium blue reagent soln. [30] and heated to 100° for 3 min. After cooling to 25°, the rate of hydrolysis was determined by measuring the absorption at $\lambda = 660$ nm and subsequently subtracting the absorption of a blank probe (H₂O, buffer, substrate).

e) *Inhibition of Snail β -Mannosidase*. $K_M = 0.58$ mM ([10]; $K_M = 0.42$ –0.80 mM). The inhibition studies were carried out at pH 4.5 (AcONa buffer, 0.04M) and 25°. The reaction was started by the addition of 4-nitrophenyl β -D-mannopyranoside after pre-incubating the enzyme in the presence of the inhibitor for 120 min at 25°. After 5 min, the enzyme reaction was quenched by the addition of borate buffer (pH 9.2, 0.2M). The rate of hydrolysis was determined by measuring the absorption at $\lambda = 405$ nm and subsequently subtracting the absorption of a blank probe (H₂O, buffer, substrate).

REFERENCES

- [1] P. Deslongchamps, 'Stereo-electronic Effects in Organic Chemistry', Pergamon Press, Oxford, 1983; A. J. Kirby, 'The Anomeric Effect and Related Stereo-electronic Effects at Oxygen', Springer-Verlag, Berlin, 1983; P. Deslongchamps, *Pure Appl. Chem.* **1993**, *65*, 1161.
- [2] A. Vasella, G. J. Davies, M. Böhm, *Curr. Opin. Chem. Biol.* **2002**, 619.
- [3] P. J. Berti, H. S. E. Tanaka, *Adv. Phys. Org. Chem.* **2002**, *37*, 239; S. G. Withers, *Carbohydr. Polym.* **2001**, *44*, 325; D. L. Zechel, S. G. Withers, *Acc. Chem. Res.* **2000**, *33*, 11; C. S. Rye, S. G. Withers, *Curr. Opin. Chem. Biol.* **2000**, *4*, 573; T. D. Heightman, A. T. Vasella, *Angew. Chem., Int. Ed.* **1999**, *38*, 750; H. Ly, S. G. Withers, *Annu. Rev. Biochem.* **1999**, *68*, 487; D. L. Zechel, S. G. Withers, in 'Comprehensive Natural Product Chemistry', Ed. C. D. Poulter, Elsevier, New York, 1999, Vol. 5, p. 279; G. Davies, M. L. Sinnott, S. G. Withers, in 'Comprehensive Biological Catalysis', Academic Press, 1998, Vol. 1, p. 119; A. White, D. R. Rose, *Curr. Opin. Biol. Chem.* **1997**, *7*, 645; G. Davies, B. Henrissat, *Structure* **1995**, *3*, 853; J. D. McCarter, S. G. Withers, *Curr. Opin. Struct. Biol.* **1994**, *4*, 885.
- [4] A. E. Stütz, 'Iminosugars as Glycosidase Inhibitors: Nojirimycin and Beyond', Wiley-VCH, Weinheim, 1999.
- [5] P. M. Coutinho, B. Henrissat, in 'Recent Advances in Carbohydrate Bioengineering', Eds. H. J. Gilbert, G. Davies, B. Henrissat, and B. Svensson, The Royal Society of Chemistry, Cambridge, 1999, p. 3.
- [6] V. H. Lillelund, H. H. Jensen, X. Liang, M. Bols, *Chem. Rev.* **2002**, *102*, 515; E. S. H. El Ashry, N. Rashed, A. H. S. Shobier, *Pharmazie* **2000**, *55*, 251, 331, 403; N. Asano, *J. Enzym. Inhib.* **2000**, *15*, 215; N. Asano, R. J. Nash, R. J. Molyneux, G. W. J. Fleet, *Tetrahedron: Asymmetry* **2000**, *11*, 1645; G. Legler, *Adv. Carbohydr. Chem. Biochem.* **1990**, *48*, 319.
- [7] M. N. Namchuk, S. G. Withers, *Biochemistry* **1995**, *34*, 16194.
- [8] K.-R. Roeser, G. Legler, *Hoppe-Seyler's Z. Physiol. Chem.* **1980**, *361*, 321; K.-R. Roeser, G. Legler, *Biochim. Biophys. Acta* **1981**, *657*, 321.
- [9] N. Panday, Y. Canac, A. Vasella, *Helv. Chim. Acta* **2000**, *83*, 58.
- [10] M. Terinek, A. Vasella, *Helv. Chim. Acta* **2003**, *86*, 3482.
- [11] T. D. Heightman, A. Vasella, K. E. Tsitsanou, S. E. Zographos, V. T. Skamnaki, N. G. Oikonomakos, *Helv. Chim. Acta* **1998**, *81*, 853.
- [12] M. Böhm, E. Lorthiois, M. Meyyappan, A. Vasella, *Helv. Chim. Acta* **2003**, *86*, 3787.
- [13] V. M.-A. Ducros, D. L. Zechel, G. N. Murshudov, H. J. Gilbert, L. Szabó, D. Stoll, S. G. Withers, G. J. Davies, *Angew. Chem., Int. Ed.* **2002**, *41*, 2824.
- [14] E. Lorthiois, M. Meyyappan, A. Vasella, *Chem. Commun.* **2000**, 1829.
- [15] Y. Kishi, M. Aratani, H. Tanino, T. Fukuyama, T. Goto, S. Inoue, S. Sugiura, H. Kakoi, *J. Chem. Soc., Chem. Commun.* **1972**, 64.

- [16] J. P. Kutney, G. H. Bokelman, M. Ichikawa, E. Jahngen, A. V. Joshua, P.-H. Liao, B. R. Worth, *Can. J. Chem.* **1977**, *55*, 3227.
- [17] M. Kawana, S. Emoto, *Tetrahedron Lett.* **1975**, *39*, 3395.
- [18] A. R. Plant, J. E. Oliver, M. L. Patchett, R. M. Daniel, H. W. Morgan, *Arch. Biochem. Biophys.* **1988**, *262*, 181.
- [19] G. Legler, *Biochim. Biophys. Acta* **1978**, *524*, 94.
- [20] M. Kleban, P. Hilgers, J. N. Greul, R. D. Kugler, J. Li, S. Picasso, P. Vogel, V. Jäger, *ChemBioChem* **2001**, *2*, 365.
- [21] G. M. Sheldrick, 'SHELXS-96, Program for the Solution of Crystal Structures'. Universität Göttingen, Germany, 1996.
- [22] G. M. Sheldrick, 'SHELXL-97, Program for the Refinement of Crystal Structures'. Universität Göttingen, Germany, 1997.
- [23] C. K. Johnson, 'ORTEP II, A Fortran Thermal-Ellipsoid Plot Program. Report ORNL-5138', Oak Ridge National Laboratory, Oak Ridge, Tennessee, USA, 1976.
- [24] W. Adam, J. Bialas, L. Hadjarapoglou, *Chem. Ber.* **1991**, *124*, 2377.
- [25] A. Altomare, B. Carrozzini, G. L. Cascarano, C. Giacovazzo, A. Guagliardi, A. G. Moliterni, R. Rizzi, 'A package for crystal structure solution by direct methods and refinement', Istituto di Ricerca per lo Sviluppo di Metodologie Cristallografiche, CNR, Campus Universitario, Via Orabona 4, 70125 Bari, Italia, 1997.
- [26] S. Aoyagi, R. Tanaka, M. Naruse, C. Kibayashi, *J. Org. Chem.* **1998**, *63*, 8397.
- [27] I. H. Segel, 'Enzyme Kinetics', Wiley, New York, 1975.
- [28] C. Blüchel, C. V. Ramana, A. Vasella, *Helv. Chim. Acta* **2003**, *86*, 2998.
- [29] S. Vonhoff, K. Piens, M. Pipelier, C. Braet, M. Claeysens, A. Vasella, *Helv. Chim. Acta* **1999**, *82*, 963.
- [30] C. K. Jue, P. N. Lipke, *J. Biochem. Biophys. Methods* **1985**, *11*, 109.

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